

Atomic Number

- Increase across a period
- increase down a family

Atomic Mass

- increase across a period
- increase down a family

Atomic Radius

half the distance between nuclei of two like atoms.

- decreases across a period (due to increased attraction of more positive nuclei to electron clouds)
- increase down a family (due to addition of an energy level)

Electron Affinity

amount of energy when an atom gains an electron

- increases across a period
- decreases down a family

Ionization Energy

amount of energy required to remove an electron from a neutral atom

- increase across a period
- decrease down a family

Electron Distribution - Forming Ions

- Atoms become ions by gaining or losing electrons to become an ion.
- An atom is stable when it has a full outer energy level (8 valence electrons).
- Atoms that lose (give away) electrons become positive charged ions. Atoms that gain (steal) electrons become negative charged ions.

Note Taking Guide Chemistry Episode 403

J Elliott



Note Taking Guide Chemistry Episode 403:

Decoding **Note Taking Guide Chemistry Episode 403**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Note Taking Guide Chemistry Episode 403**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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