

Background

- 1. Study of this _____ multi _____ relationship is a _____ situation.
- 2. Research on _____ relationship _____ equities.

2. $40g = 40r + 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide.

The _____ is a _____ reaction _____ the _____ _____ _____ from the _____ reaction in the _____.

3. Problem

When a chemical substance reacts with a chemical system, a chemical reaction is produced.

$$2 \text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

moles of Al

$$\begin{array}{ccc} \frac{1}{2} & \text{Al} & \frac{3}{2} & \text{O}_2 \\ \frac{1}{2} & \text{Al} & \frac{3}{2} & \text{Al}_2\text{O}_3 \\ \frac{1}{2} & \text{O}_2 & \frac{3}{2} & \text{Al}_2\text{O}_3 \end{array}$$

2. If you start with 4 moles of Al, how many moles of Al_2O_3 will be produced?

$$2 \text{ moles Al} \rightarrow 1 \text{ mole Al}_2\text{O}_3 \text{ and } 40g = 4 \text{ moles Al}_2\text{O}_3$$

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Padhraic Smyth



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