

A chemical equation is a representation of a chemical reaction in which the reactants are written using chemical formulas, followed by an arrow.

Reactants Products

A coefficient represents the number of atoms or molecules participating in a chemical reaction. A coefficient is placed directly in front of a chemical formula. If there is only one molecule of a substance, the coefficient is not written.

A subscript is placed after the element symbol to indicate the number of atoms of that element in a molecule. A subscript is not written if there is only one atom of an element.

Why is it important to balance chemical equations?

Chemical reactions are governed by the law of conservation of mass. This means that the total mass of the reactants must equal the total mass of the products. Balancing chemical equations ensures that the number of atoms of each element is the same on both sides of the equation.

What does the total law of conservation of mass mean when balancing equations?

The total mass of the reactants must equal the total mass of the products.

Week 8, Day 2

The number of protons in an atom cannot change (it is fixed) but a different number of neutrons can change.

The number of protons in an atom is fixed, but the number of neutrons can change. This is because protons and neutrons are made of quarks, and quarks can be created or destroyed in certain types of nuclear reactions.

The most stable atomic configuration has a nucleus with a balanced number of protons and neutrons. If there are too many protons, the nucleus will be unstable and will decay into a more stable configuration.

How does a positive ion form from a neutral atom? What is another name for a positive ion?

A positive ion forms when an atom loses one or more electrons. This is called oxidation. The atom is now positively charged and is called a cation.

How does a negative ion form from a neutral atom? What is another name for a negative ion?

A negative ion forms when an atom gains one or more electrons. This is called reduction. The atom is now negatively charged and is called an anion.

Atoms are made of protons, neutrons, and electrons. Protons are positively charged, neutrons are neutral, and electrons are negatively charged.

Atoms are made of protons, neutrons, and electrons. Protons are positively charged, neutrons are neutral, and electrons are negatively charged. In a neutral atom, the number of protons equals the number of electrons.

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Atoms are made of protons, neutrons, and electrons. Protons are positively charged, neutrons are neutral, and electrons are negatively charged.



When an atom loses one or more electrons, it becomes a positive ion. This is called oxidation. The atom is now positively charged and is called a cation.

The positively charged cation and the negatively charged anion are attracted to each other. This is called an ionic bond.

Physical Science Exploring Space Notetaking Guide

Answer

DP Hallahan



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