

Common Polyatomic Ions

| | |
|----------------------|---------------------------------------|
| acetate | $\text{C}_2\text{H}_3\text{O}_2^-$ |
| ammonium | NH_4^+ |
| arsenate | AsO_4^{3-} |
| arsenite | AsO_3^{3-} |
| azide | N_3^- |
| benzoate | $\text{C}_7\text{H}_5\text{O}_2^-$ |
| borate | BO_3^{3-} |
| bromate | BrO_3^- |
| carbonate | CO_3^{2-} |
| chlorate | ClO_3^- |
| chlorite | ClO_2^- |
| chromate | CrO_4^{2-} |
| cyanide | CN^- |
| dichromate | $\text{Cr}_2\text{O}_7^{2-}$ |
| dihydrogen phosphate | H_2PO_4^- |
| dihydrogen phosphite | H_2PO_3^- |
| hydrogen carbonate | HCO_3^- |
| hydrogen phosphate | HPO_4^{2-} |
| hydrogen phosphite | HPO_3^{2-} |
| hydrogen sulfate | HSO_4^- |
| hydrogen sulfide | HS^- |
| hydrogen sulfite | HSO_3^- |
| hydroxide | OH^- |
| hypochlorite | ClO^- |
| iodate | IO_3^- |
| manganate | MnO_4^{2-} |
| nitrate | NO_3^- |
| nitrite | NO_2^- |
| oxalate | $\text{C}_2\text{O}_4^{2-}$ |
| perchlorate | ClO_4^- |
| permanganate | MnO_4^- |
| peroxide | O_2^{2-} |
| phosphate | PO_4^{3-} |
| phosphite | PO_3^{3-} |
| silicate | SiO_3^{2-} |
| sulfate | SO_4^{2-} |
| sulfite | SO_3^{2-} |
| tartrate | $\text{C}_4\text{H}_4\text{O}_6^{2-}$ |
| thiocyanate | SCN^- |
| thiosulfate | $\text{S}_2\text{O}_3^{2-}$ |

| | |
|---------------------------------------|----------------------|
| AsO_3^{3-} | arsenite |
| AsO_4^{3-} | arsenate |
| BO_3^{3-} | borate |
| BrO_3^- | bromate |
| $\text{C}_2\text{H}_3\text{O}_2^-$ | acetate |
| $\text{C}_2\text{O}_4^{2-}$ | oxalate |
| $\text{C}_4\text{H}_4\text{O}_6^{2-}$ | tartrate |
| $\text{C}_7\text{H}_5\text{O}_2^-$ | benzoate |
| ClO^- | hypochlorite |
| ClO_2^- | chlorite |
| ClO_3^- | chlorate |
| ClO_4^- | perchlorate |
| CN^- | cyanide |
| CO_3^{2-} | carbonate |
| $\text{Cr}_2\text{O}_7^{2-}$ | dichromate |
| CrO_4^{2-} | chromate |
| H_2PO_3^- | dihydrogen phosphite |
| H_2PO_4^- | dihydrogen phosphate |
| HCO_3^- | hydrogen carbonate |
| HPO_3^{2-} | hydrogen phosphite |
| HPO_4^{2-} | hydrogen phosphate |
| HS^- | hydrogen sulfide |
| HSO_3^- | hydrogen sulfite |
| HSO_4^- | hydrogen sulfate |
| IO_3^- | iodate |
| MnO_4^- | permanganate |
| MnO_4^{2-} | manganate |
| N_3^- | azide |
| NH_4^+ | ammonium |
| NO_2^- | nitrite |
| NO_3^- | nitrate |
| O_2^{2-} | peroxide |
| OH^- | hydroxide |
| PO_3^{3-} | phosphite |
| PO_4^{3-} | phosphate |
| $\text{S}_2\text{O}_3^{2-}$ | thiosulfate |
| SCN^- | thiocyanate |
| SiO_3^{2-} | silicate |
| SO_3^{2-} | sulfite |
| SO_4^{2-} | sulfate |

Physical Science Formulas With Polyatomic Ions Answer

Gary Raham



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according to lopez the lotus is clearly a work of high literary quality its authors are unknown but they were likely highly educated buddhist monks fully at ease among the doctrines and tropes of buddhism as it existed in india at the time according to peter alan roberts the lotus sūtra may have had its origin among the mahāsāṃghika school and may have been written in a middle indic language

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