

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
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Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																									
speed of light in vacuum																		c		299 792 458 m s ⁻¹ (exact)					
Planck constant																		h		6.626 070 15 × 10 ⁻³⁴ J s (exact)					
elementary charge																		e		1.602 176 634 × 10 ⁻¹⁹ C					
electron mass																		m _e		9.109 383 56 × 10 ⁻³¹ kg					
																		m _e /m _p		5.485 799 094 6 × 10 ⁻⁴					
proton mass																		m _p		1.672 621 9 × 10 ⁻²⁷ kg					
fine-structure constant																		α		1/137.035 999 084					
Rydberg constant																		R _∞		10 973 731 568 162 854 m ⁻¹					
																		R _H		10 967 732 320 869 896 m ⁻¹					
																		R _{He}		10 997 932 170 746 876 m ⁻¹					
Schrödinger constant																		k		1.380 658 × 10 ⁻²³ J K ⁻¹					

Atomic Number: 58
Ground-state Ionization Energy (eV): 5.5357

Symbol: **Ce**
Name: Cerium
Atomic Weight: 140.116
Ground-state Configuration: [Xe]4f¹5d¹6s²

*Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

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David Baud

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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