

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																									
speed of light in vacuum c 299 792 458 m s ⁻¹ (exact)																									
Planck constant h 6.626 070 15 × 10 ⁻³⁴ J s (h = h/2π)																									
elementary charge e 1.602 176 634 × 10 ⁻¹⁹ C																									
electron mass m _e 9.109 383 56 × 10 ⁻³¹ kg																									
proton mass m _p 1.672 621 9 × 10 ⁻²⁷ kg																									
fine-structure constant α 1/137.035 999 073 92																									
Rydberg constant R _∞ 10 973 731.568 162 85 × 10 ¹⁰ m ⁻¹																									
Boltzmann constant k 1.380 658 × 10 ⁻²³ J K ⁻¹																									

Atomic Number: 58
Ground-state Ionization Energy (eV): 5.5337

Symbol: **Ce**
Name: Cerium
Atomic Weight: 140.116

Ground-state Configuration: [Xe]4f¹5d¹6s²

*Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2010)

Periodix Table Study Packet

Peter L. Kresan, Reed Mencke

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the authors' names.

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