| energy:   | he element in each of the t  |  |
|---|--|--|
| a) lithium  | or boron   | c) cesium or lead  |
| b) magnes   | ium or strontium   | d) sodium or argon   |
| 1. Arrange the  | e following elements in ord  | ler of increasing ionization energy:   |
| a) Sr, Be, A  | Mg   |  |
| b) Bi, Cs,  | Ba   |  |
| c) Na, Al,  | s  |  |
| 2. Why is ther elements in  |  | ween the 1st and 2nd ionization energies for the   |
| Name:   |  | Date: Class period:  |
|   | "lonic Size ar   | Periodic Trends<br>nd E.N. Trend" - WS #4  |
| Pirections: Pla   | ease answer each fill in the   | e blank with the best answer.     O  |
| . A cation is   | an ion with an overall   | charge.  |
|   | The second secon |  |
| . Cations are   | made when an atom  | <u>@</u>   |
|   |  | than the size of the neutral   |
| . The size of   |  |  |
| . The size of atom of the   | a positive ion is always<br>e same element.  |  |
| . The size of<br>atom of the<br>. As the posi   | a positive ion is always<br>e same element.<br>tive charge on an ion incre   | than the size of the neutral   |
| . The size of<br>atom of the<br>. As the posi<br>. An anion is  | a positive ion is always<br>e same element.<br>tive charge on an ion incre   | ases, the size of the atom will charge.  |
| . The size of<br>atom of the<br>. As the posi<br>. An anion is<br>. Anions are  | a positive ion is always<br>e same element.<br>tive charge on an ion incre<br>an ion with an overall<br>made when at atom  | ases, the size of the atom will charge.  |
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| . The size of atom of the . As the position . An anion is . Anions are . The size of atom of the . As the negative . Which elements . Which elements .  | a positive ion is always e same element.  tive charge on an ion incre an ion with an overall made when at atom a negative ion is always e same element.  ative charge on an ion incre nent/ion has the biggest at answer.  Na*1  | ases, the size of the atom will charge than the size of the neutral charge of the neutral charge of the neutral charge of the neutral charge of the size of the neutral charge of the size of the following pairs? Please comic/ionic size in each of the following pairs? Please comic/ionic size in each of the following pairs? |
| . The size of atom of the . As the position . An anion is . Anions are . The size of atom of the . As the negative . Which elemnicircle your a) Na or I | a positive ion is always e same element.  tive charge on an ion incre an ion with an overall made when at atom a negative ion is always e same element.  ative charge on an ion incre nent/ion has the biggest at answer.  Na*1  | than the size of the neutral ases, the size of the atom will charge than the size of the neutral eases, the size of the atom will comic/ionic size in each of the following pairs? Please d) Al or Al -3   |



# **Periodic Trends Unit Exam And Answers**

**Mike Jess** 

**Periodic Trends Unit Exam And Answers:** 

## Periodic Trends Unit Exam And Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has are more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Periodic Trends Unit Exam And Answers**, a literary masterpiece that delves deep in to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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