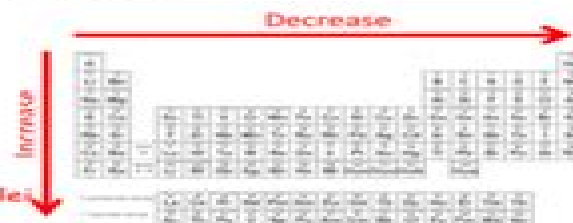


Periodic Trends Test Review

Atomic Radii (Atomic Size)

- Draw the trend lines for atomic radius on the periodic table to the right.
- What is the definition of atomic radius?
Atomic radius is one half the distance between the nuclei of two atoms (of the same element) that are bonded together.
- What happens to the size of atoms as you move down a group? **Increases**
Why? **Down a group each element has an additional (and larger) energy level to hold its electrons.**
- What happens to the size of an atom as you move left to right across a period? **Decreases**
Why? **Across a period each successive element has an additional proton which increases the positive charge in the nucleus. The positive charge attracts the negative electrons, and pulls them toward the nucleus, making the atom smaller.**
- How are the numbers of valence electrons of the Group A elements related to the group number? **The group number tells you how many valence electrons elements in that group have. For example, Group 1A elements each have 1 valence electron.**
- How is the energy level of an atom's valence electrons related to the period it is in on the periodic table? **The period number tells you the energy level that an atom's valence electron(s) is/are located. For example, Sodium (Na) is in the third period. Its valence electron is located in the 3rd energy level.**
- Put the following sets of elements in order of **increasing** atomic size (smallest to largest).

a. Aluminum, Sodium, Chlorine	Cl, Al, Na
b. Rubidium, Lithium, Potassium	Li, K, Rb
c. Tin, Iodine, Xenon	Xe, I, Sn
d. Bismuth, Phosphorous, Nitrogen	N, P, Bi
e. Beryllium, Barium, Calcium	Be, Ca, Ba
- Given any two elements within a group, is the element with the smaller or larger atomic number likely to have a larger or smaller atomic radius than the other element?
The element with the larger atomic number will have the larger atomic radius.



Periodic Trends Test Review Answers

R Bogdan



Periodic Trends Test Review Answers:

Periodic Trends Test Review Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Periodic Trends Test Review Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Table of Contents Periodic Trends Test Review Answers

1. Understanding the eBook Periodic Trends Test Review Answers
 - The Rise of Digital Reading Periodic Trends Test Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Test Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Test Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Test Review Answers
 - Personalized Recommendations
 - Periodic Trends Test Review Answers User Reviews and Ratings
 - Periodic Trends Test Review Answers and Bestseller Lists

5. Accessing Periodic Trends Test Review Answers Free and Paid eBooks
 - Periodic Trends Test Review Answers Public Domain eBooks
 - Periodic Trends Test Review Answers eBook Subscription Services
 - Periodic Trends Test Review Answers Budget-Friendly Options
6. Navigating Periodic Trends Test Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Test Review Answers Compatibility with Devices
 - Periodic Trends Test Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Test Review Answers
 - Highlighting and Note-Taking Periodic Trends Test Review Answers
 - Interactive Elements Periodic Trends Test Review Answers
8. Staying Engaged with Periodic Trends Test Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Test Review Answers
9. Balancing eBooks and Physical Books Periodic Trends Test Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Test Review Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Test Review Answers
 - Setting Reading Goals Periodic Trends Test Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Test Review Answers
 - Fact-Checking eBook Content of Periodic Trends Test Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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