

Name: \_\_\_\_\_ ( ) Class: \_\_\_\_\_ Date: \_\_\_\_\_

The Periodic TableGeneral Properties of the Periodic Table

1. The Periodic Table is a table that groups elements according to certain physical properties or chemical properties.
2. The elements are arranged in order of increasing proton number (atomic number).
3. A **HORIZONTAL** row of elements is known as a Period.  
Across a period, the element changes from **metal** character to **non-metal** character.  
Elements on left hand side of Periodic Table are \_\_\_\_\_.  
Elements on right hand side of Periodic Table are \_\_\_\_\_.
4. A **VERTICAL** row of elements is known as a Group.
5. Hydrogen and Helium are in the same Period 1.  
They all have \_\_\_\_\_ electron shell.

Draw out the electronic structure of Hydrogen and Helium below:



Hydrogen



Helium

6. The following elements are in the same Period 2.  
They all have \_\_\_\_\_

Draw out the electronic structure of these elements.

--	--	--	--	--	--	--	--

Lithium

Beryllium

Boron

Carbon

Nitrogen

Oxygen

Fluorine

Neon

7. Lithium is in Group 1. Lithium has \_\_\_\_\_ outer electron.  
Beryllium is in Group 2. Beryllium has \_\_\_\_\_ outer electrons.  
Oxygen is in Group 6. Oxygen has \_\_\_\_\_

Quick Check

Without reference to the Periodic Table, predict the Period and Group of the elements below:

- |                  |       |                      |       |
|------------------|-------|----------------------|-------|
| a) Ca – 2.8.8.2  | _____ | d) Xe 2.8.18.18.8    | _____ |
| b) S – 2.8.6     | _____ | e) Pb 2.8.18.32.18.4 | _____ |
| c) Ga – 2.8.18.3 | _____ | f) Rb – 2.8.18.8.1   | _____ |

# Periodic Table The Science Spot Answer Key

**Wolfgang Guggemos**



## **Periodic Table The Science Spot Answer Key:**

*Spotlight Science* Keith Johnson, Sue Adamson, Gareth Williams, 2002 This Spiral Edition Teacher Support Pack offers comprehensive support and guidance providing the best possible learning experience for your students and saving time for everyone in the department

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**Table of Contents Periodic Table The Science Spot Answer Key**

1. Understanding the eBook Periodic Table The Science Spot Answer Key
  - The Rise of Digital Reading Periodic Table The Science Spot Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table The Science Spot Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table The Science Spot Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table The Science Spot Answer Key
  - Personalized Recommendations
  - Periodic Table The Science Spot Answer Key User Reviews and Ratings
  - Periodic Table The Science Spot Answer Key and Bestseller Lists
5. Accessing Periodic Table The Science Spot Answer Key Free and Paid eBooks
  - Periodic Table The Science Spot Answer Key Public Domain eBooks
  - Periodic Table The Science Spot Answer Key eBook Subscription Services
  - Periodic Table The Science Spot Answer Key Budget-Friendly Options
6. Navigating Periodic Table The Science Spot Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table The Science Spot Answer Key Compatibility with Devices
  - Periodic Table The Science Spot Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table The Science Spot Answer Key
  - Highlighting and Note-Taking Periodic Table The Science Spot Answer Key
  - Interactive Elements Periodic Table The Science Spot Answer Key
8. Staying Engaged with Periodic Table The Science Spot Answer Key



- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table The Science Spot Answer Key
9. Balancing eBooks and Physical Books Periodic Table The Science Spot Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table The Science Spot Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table The Science Spot Answer Key
- Setting Reading Goals Periodic Table The Science Spot Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table The Science Spot Answer Key
- Fact-Checking eBook Content of Periodic Table The Science Spot Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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