Periodic Table Extra Practice Worksheet

Directions: Answer the following questions using the Periodic Table below.

IA.							VIII
1 H 1	IIA	IIIB	IVB	VB	VIB	VIIB	2 He 4
3 Li 7	Be 9	5 B 11	6 C 12	7 N 14	8 O 16	9 F 19	10 Ne 20
11 Na 23	12 Mg 24	13 Al 27	14 Si 28	15 P 31	16 S 32	17 CI 35	18 Ar 40

1. How many protons, neutrons, and electrons are in each of the following atoms?

Element	Boron (B)	Sodium (Na)	Phosphorus (P)
Protons	5	11.	15
Neutrons	6	12	16
Electrons	5	11	15

2. Which diagram best shows nucleus of beryllium? Circle the correct answer.

$$\begin{pmatrix}
n = 4 \\
e = 5
\end{pmatrix}$$
 $\begin{pmatrix}
p = 4 \\
e = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 5 \\
n = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 5 \\
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\end{pmatrix}$
 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$
 $\begin{pmatrix}
p = 4 \\
n = 4
\end{pmatrix}$
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$

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3. Fill in the chart below from the data provided.

ELEMENT	ATOMIC NUMBER	MASS NUMBER	NUMBER OF PROTONS	NUMBER OF NEUTRONS	NUMBER OF ELECTRONS
Aluminum	a.13	27	b.13	c.14	d.13
Hydrogen	e.1	f.1	1	g.0	h.1
Argon	i.18	j.40	k.18	1.22	18
Magnesium	12	m.24	n.12	0.12	p.12

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SA Adler

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Table of Contents Periodic Table Practice Answers

- 1. Understanding the eBook Periodic Table Practice Answers
 - The Rise of Digital Reading Periodic Table Practice Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Practice Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Practice Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Table Practice Answers
 - Personalized Recommendations

- Periodic Table Practice Answers User Reviews and Ratings
- Periodic Table Practice Answers and Bestseller Lists
- 5. Accessing Periodic Table Practice Answers Free and Paid eBooks
 - o Periodic Table Practice Answers Public Domain eBooks
 - Periodic Table Practice Answers eBook Subscription Services
 - Periodic Table Practice Answers Budget-Friendly Options
- 6. Navigating Periodic Table Practice Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Periodic Table Practice Answers Compatibility with Devices
 - Periodic Table Practice Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Periodic Table Practice Answers
 - Highlighting and Note-Taking Periodic Table Practice Answers
 - Interactive Elements Periodic Table Practice Answers
- 8. Staying Engaged with Periodic Table Practice Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Practice Answers
- 9. Balancing eBooks and Physical Books Periodic Table Practice Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Practice Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Practice Answers
 - Setting Reading Goals Periodic Table Practice Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Practice Answers
 - Fact-Checking eBook Content of Periodic Table Practice Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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