## Periodic Table Notes: Determining # of Shells and Valence Electrons

#### Directions:

View the Power Point study guide with the same title as this handout. Record all your answers below.

- 1. In the periodic table, each row is called a \_\_\_\_\_\_.
- 2. The elements in each \_\_\_\_\_\_ have the same number of
- 3. Each column in the periodic table is called a \_\_\_\_\_\_.
- 4. Each of the elements in the same \_\_\_\_\_ have the same number of \_\_\_\_ in their outer shells.
- 5. The electrons in the outer shell are called \_\_\_\_\_\_ electrons.
- 6. The \_\_\_\_\_ number equals the \_\_\_\_\_ of \_\_\_\_\_electrons.

Complete the following data table as you view the rest of the slides.

Symbol	Element	Period	# of Shells	Group #	# Valence electrons
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