

Chapter 5 The Periodic Table

Section 5.2 The Modern Periodic Table

(pages 130–138)

This section explains the organization of the modern periodic table and discusses the general properties of metals, nonmetals, and metalloids.

Reading Strategy (page 130)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132–133. As you read, write answers to your questions. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Questions About the Periodic Table	
Question	Answer
Students might ask: What does atomic mass mean?	Atomic mass is a value that depends on the distribution of an element's isotopes in nature and the masses of those isotopes.
Why are two columns of elements placed before the main body of the table?	Two kinds of elements are placed before the table to make the table more compact.

Other possible questions: Why are there two numbering systems for the columns? Why is period 7 incomplete?

The Periodic Law (pages 131–133)

1. Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.
_____ **true** _____
2. Explain why the number of elements per period varies. _____ **The number of elements per period varies because the number of available orbitals increases from energy level to energy level.** _____
3. Properties of elements repeat in a predictable way when atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the _____ **periodic law** _____.

Atomic Mass (page 134)

4. Label the four types of information supplied for chlorine in the diagram.

a.	17
b.	Cl
c.	Chlorine
d.	35.453

- Atomic number**
- Element symbol**
- Element name**
- Atomic mass**

Periodic Table Section 5 2 Answer

DJ Losen



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Physical Chemistry Peter Atkins, Julio de Paula, Ronald Friedman, 2014 This title takes an innovative molecular approach to the teaching of physical chemistry The authors present the subject in a rigorous but accessible manner allowing students to gain a thorough understanding of physical chemistry

Official Gazette of the United States Patent and Trademark Office United States. Patent and Trademark Office, 2000

Essential AS Chemistry for OCR Ted Lister, Janet Renshaw, 2004 Essential AS Chemistry for OCR provides clear progression with challenging material for in depth learning and understanding Written by the best selling authors of New Understanding Chemistry these texts have been written in simple easy to understand language and each double page spread is designed in a contemporary manner Fully networkable and editable Teacher Support CD ROMs are also available for this series they contain worksheets marking schemes and practical help

The Sciences James Trefil, Robert M. Hazen, 2016-09-13 This text is an unbound three hole punched version The Sciences An Integrated Approach Binder Ready Version 8th Edition by James Trefil and Robert Hazen uses an approach that recognizes that science forms a seamless web of knowledge about the universe This text fully integrates physics chemistry astronomy earth sciences and biology and emphasizes general principles and their application to real world situations The goal of the text is to help students achieve scientific literacy Applauded by students and instructors for its easy to read style and detail appropriate for non science majors the eighth edition has been updated to bring the most up to date coverage to the students in all areas of science

International Symposium on Computer and Information Sciences Erol Gelenbe, 2022-09-16 Papers from an October 2002 symposium describe research in areas including algorithms artificial intelligence computer graphics computer networks databases evolutionary computation graph theory image processing multimedia technology software engineering and software performance engineering

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