

Name: _____

Atomic Structure & the Periodic Table Study Guide

Atomic Structure

Particle	Charge	Where are they found?
Proton	Positive	Nucleus
Neutron	Neutral/ no charge	Nucleus
Electron	negative	Electron cloud/energy levels outside nucleus

1. What is the rule for writing chemical symbols like Na or Ca? **Capitalize the first letter; lower case second letter**
2. What are the vertical columns on the periodic table called? Groups or families
 - a. How many of these are on the periodic table? 18
 - b. Why are the elements in these columns? They have the same chemical properties (& same # of valence electrons)
 - c. Elements in the same group have the same number of valence electrons.
3. What are the horizontal rows on the periodic table called? periods
 - a. How many of these are on the periodic table? 7
 - b. Elements in these rows have the same number of energy (*electron shells*) levels.
4. What are valence electrons? **Electrons in the outer energy level**
5. How many electrons are in each energy level?
1st energy level 2 2nd energy level 8 3rd energy level 8
6. The atomic number is equal to the # of protons and also # of electrons
7. The atomic mass is equal to the # protons + # of neutrons
8. How do you find the number of neutrons? Atomic mass – atomic number
9. We view the order of the periodic table today according to the atomic number

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Table of Contents Periodic Table Printable Study Guide

1. Understanding the eBook Periodic Table Printable Study Guide
 - The Rise of Digital Reading Periodic Table Printable Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Printable Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Printable Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Printable Study Guide
 - Personalized Recommendations
 - Periodic Table Printable Study Guide User Reviews and Ratings

- Periodic Table Printable Study Guide and Bestseller Lists
- 5. Accessing Periodic Table Printable Study Guide Free and Paid eBooks
 - Periodic Table Printable Study Guide Public Domain eBooks
 - Periodic Table Printable Study Guide eBook Subscription Services
 - Periodic Table Printable Study Guide Budget-Friendly Options
- 6. Navigating Periodic Table Printable Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Printable Study Guide Compatibility with Devices
 - Periodic Table Printable Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Printable Study Guide
 - Highlighting and Note-Taking Periodic Table Printable Study Guide
 - Interactive Elements Periodic Table Printable Study Guide
- 8. Staying Engaged with Periodic Table Printable Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Printable Study Guide
- 9. Balancing eBooks and Physical Books Periodic Table Printable Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Printable Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Printable Study Guide
 - Setting Reading Goals Periodic Table Printable Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Printable Study Guide
 - Fact-Checking eBook Content of Periodic Table Printable Study Guide
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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