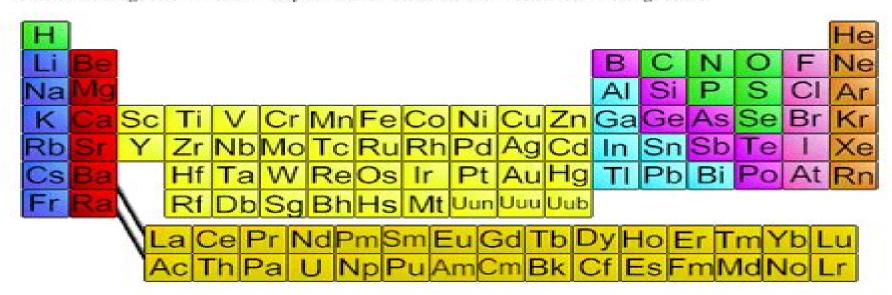


Part 1:

Follow the instructions in the Periodic Table Virtual Lab Lesson to color the periodic table.

Click on the image below -> edit -> drop down arrow beside the line -> select tool -> change colors.



Part 2: Go to chemicalelements com.

- 1. Click on Alkali Metals (left bar) to answer the following questions:
 - a. What is the group number? [1]
 - b. Are these metals reactive? [Yes, they are highly reactive.]
 - Do these metals occur freely in nature? [No, they don't occur freely in nature, but they are found in various minerals.]
 - d. How many electrons are in their outer shell? [1]
 - What are the three characteristics of ALL metals? [Shiny luster, good conductors of heat and electricity, malleable and ductile.
 - f. Are these metals soft or hard? [soft]
 - g. Name the two most reactive elements in this group? [Potassium and Sodium]
 - h. What happens when they are exposed to water? [They react vigorously with water, producing hydrogen gas and an alkaline solution. The reaction can be violent and can even ignite the hydrogen gas produced...]

Periodic Table Grouping Lab Answer Key

Annelies Wilder-Smith

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