# Percent Composition

So the percentage of carbon in ethane is...C<sub>2</sub>H<sub>6</sub>

$$%C = \frac{(2)(12.0 \text{ amu})}{(30.0 \text{ amu})}$$

$$= \frac{24.0 \text{ amu}}{30.0 \text{ amu}} \times 100$$

$$= 80.0\%$$



# **Percentage Composition Answers Prentice Hall**

**Michael Brown** 

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