

Percent Yield Homework

I performed the following chemical reaction:



- 1) Balance the equation in the spaces above.
- 2) What type of reaction is taking place here? _____
- 3) Complete the equation by showing symbols of state as well as symbols around the arrow and H₂ values, if possible.
- 4) If I were to start with 40 grams of $\text{C}_2\text{H}_5\text{OH}$ and an excess of oxygen, how many grams of carbon dioxide might I expect to make?
- 5) If my actual yield of carbon dioxide was 25 grams, what is my percent yield for this reaction?
- 6) Is my answer from problem #4 reasonable? If it is, explain why. If not, explain why not.
- 7) Explain why chemical reactions very rarely have a percent yield of 100%.

Percent Yield Sheet

**Daniel Guggenheim Aeronautical
Laboratory**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the text box.

Percent Yield Sheet:

Percent Yield Sheet Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Percent Yield Sheet**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall affect on readers.

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