



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

Duration: 120
Time: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.
Use permanent marker for writing and drawing.

QUESTION 3

PHYSICAL SCIENCES PAPER 2

INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

Examiner's use: _____ Physical Sciences P2

QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the ΔH_{vap} and ΔH_{fusion} for 1-hexanol in kJ mol⁻¹.

Compound	Molecular mass (M _r)	Boiling point (°C)
A	102	162
B	102	17

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol composed of one (1) or two (2) and 100 (one) or 50 (one) percent? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of pure compounds A and B and its boiling point of these molecules. (2)
- 3.1.4 What (100) of the compound (A or B) has a (1-hexanol) system which (propagates) bonds? (2)
- 3.1.5 Which compound (A or B) will have the higher vapour pressure? (2)
- 3.1.6 Which compound (A or B) will have the higher surface tension? (2)

Compound B (H₂O) will show in the table, has a smaller molecular mass than both compounds A and B but has a relatively higher boiling point of 100 °C.

- 3.1.6 Explain why the boiling point of compound B is (100°C) (2 marks) in comparison to A and B by referring to the (H₂O) and intermolecular or intramolecular bonds. (2)

3.2 Add table (2) to obtain a total of 10 marks. (10 marks) (10 marks)



- 3.2.1 Is solid (100) or (100), all the solid (100)? (2)
- 3.2.2 Explain the process in QUESTION 3.2.1 about the solid in the (intermolecular) bonds. (2)

Examiner's use

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Xiaolong Qi



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