EXAMINATION DATA SHEET FOR THE PHYSICAL SCIENCES (CHEMISTRY)

TABLE 1 PHYSICAL CONSTANTS

NAME	SYMBOL	VALUE
Magnitude of charge on electron	6	1,6 × 10 ⁻¹⁹ C
Mass of an electron	me	9,1 × 10 ⁻³¹ kg
Standard pressure	₽ ⁶	1,01 × 10 ⁵ Pa
Motar gas volume at STP	V.,	22,4 dm ³ - mol ⁻¹
Standard temperature	T*	273 K (0 °C)
Avogadro's constant	Na	6,02 × 10 ²³ mol ⁻¹
Faraday's constant	F	96 500 C- mol-1

TABLE 2 CHEMISTRY FORMULAE

$n = \frac{m}{M}$	$n = \frac{N}{N_A}$	$n = \frac{v}{v_{\infty}}$
$c = \frac{n}{V}$ OR $c = \frac{m}{MV}$		o*] [OHT] = 1 × 10 ⁻¹⁴ 25 °C (298 K)
q = It	$E_{cell}^{\theta} = E_{cath}^{\theta}$	ode - E ⁸ anode
q = nF	$E_{cell}^{\theta} = E_{oxidising}^{\theta}$ age	ent - E ^θ reducing agent

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