



MERCY JUNIOR COLLEGE
Grade 11
Physical Science
Finals Exam

NAME: _____

YEAR LEVEL & SECTION: _____

TEACHER'S NAME: _____

PROCTOR'S NAME: _____

1. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol e^- or β^- , with a negative elementary electric charge.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol p or p^+ , with a positive electric charge of $+1e$ elementary charge and mass slightly less than that of a neutron.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol n or n^0 , with no net electric charge and a mass slightly larger than that of a proton.
 - Neutron
 - Electron
 - Proton
 - Atom
- Is this an example of ionic bonding?



carbon dioxide

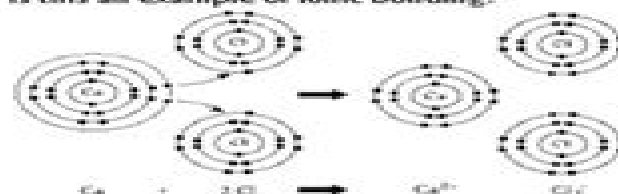
- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- Is this an example of ionic bonding?



lithium fluoride

- Yes, it is ionic bonding.
 - No, it is of another type of chemical bonding.
 - There is no chemical bonding present.
 - This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
 - Electron Configuration
 - Periodic Table of Elements
 - Orbital Models
 - Energy Levels
 - Is this an example of ionic bonding?



- Yes, it is ionic bonding.
 - No, it is of another type of chemical bonding.
 - There is no chemical bonding present.
 - This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
 - Octet Rule
 - Valence Electrons
 - Atomic Orbitals
 - Chemical Bonds
 - It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - Neutron
 - Electron
 - Proton
 - Atom
 - It is a type of chemical bond that involves the electrostatic attraction between oppositely

Physical Science Grade 11 Final Examination 2014

Scope

RD Boyd



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"Code of Massachusetts regulations, 2014", 2014 Archival snapshot of entire looseleaf Code of Massachusetts Regulations held by the Social Law Library of Massachusetts as of January 2020 Introduction to Teaching Gene E. Hall, Linda F. Quinn, Donna M. Gollnick, 2018-11-29 An ideal introductory text for aspiring teachers Introduction to Teaching Making a Difference in Student Learning is grounded in the realities and complexities found in today's schools Acclaimed authors Gene E Hall Linda F Quinn and Donna M Gollnick thoroughly prepare readers to make a difference as teachers presenting firsthand stories and evidence based practices while offering a student centered approach to learning The authors focus on how to address one of the biggest challenges facing many of today's schools ensuring that all students are learning and help teachers prioritize student learning as their primary focus From true to life challenges that future teachers will face such as high stakes testing reduced funding low retention and Common Core State Standards to the inspiration and joy they will experience throughout their teaching careers the Third Edition paints an importantly authentic picture of the real life of a teacher **Exploring Lifespan Development** Laura E. Berk, 2022-07-26 Exploring Lifespan Development Fourth Edition the essentials version of Development Through the Lifespan Seventh Edition by best selling author Laura E Berk includes the same topics the same number of chapters and the same outstanding features with a focus on the most important information and a greater emphasis on practical real life applications Applied Health Economics for Public Health Practice and Research Rhiannon Tudor Edwards, Emma McIntosh, 2019-03-14 In today's world of scarce resources determining the optimal allocation of funds to preventive health care interventions PHIs is a challenge The upfront investments needed must be viewed as long term projects the benefits of which we will experience in the future The long term positive change to PHIs from economic investment can be seen across multiple sectors such as health care education employment and beyond Applied Health Economics for Public Health Practice and Research is the fifth in the series of Handbooks in Health Economic Evaluation It presents new research on health economics methodology and application to the evaluation of public health interventions Looking at traditional as well as novel methods of economic evaluation the book covers the history of economics of public health and the economic rationale for government investment in prevention In addition it looks at principles of health economics evidence synthesis key methods of economic evaluation with accompanying case studies and much more Looking to the future Applied Health Economics for Public Health Practice and Research presents priorities for research in the field of public health economics It acknowledges the role played by natural environment in promoting better health and the place of genetics environment and socioeconomic status in determining population health Ideal for health economists public health researchers local government workers health care professionals and those responsible for health policy development Applied Health Economics for Public Health Practice and Research is an important contribution to the economic discussion of public health and resource allocation **Parliamentary Papers** Great Britain. Parliament. House of

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Sessional Papers Great Britain. Parliament. House of Commons,1908 **The City Record** New York (N.Y.),1916

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