#### Chapter 6 Chemical Bonds

# Section 6.2 Covalent Bonding (pages 165-169)

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

#### Reading Strategy (page 165)

Relating Text and Visuals As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

#### Covalent Bonds (pages 165-167)

- Describe a covalent bond. \_\_\_\_\_\_\_
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - a. electron dot b. structural formula
  - c. space-filling d. electron cloud
- 3. Describe a molecule. \_\_\_\_\_
- 4. Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus.
- 5. Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - a. zero b. one
  - c. two d. three

# Physical Science Chapter 6 Chemical Bonds Chapter Review

**Rachel Sandford** 

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Chemistry at the Frontier with Physics and Computer Science Sergio Rampino, 2022-05-16 Chemistry at the Frontier with Physics and Computer Science Theory and Computation shows how chemical concepts relate to their physical counterparts and can be effectively explored via computational tools It provides a holistic overview of the intersection of these fields and offers practical examples on how to solve a chemical problem from a theoretical and computational perspective going from theory to models methods and implementation Sections cover both sides of the Born Oppenheimer approximation nuclear dynamics and electronic structure chemical reactions chemical bonding and cover theory to practice on three related physical problems wavepacket dynamics Hartree Fock equations and electron cloud redistribution Drawing on the interdisciplinary knowledge of its expert author this book provides a contemporary guide to theoretical and computational chemistry for all those working in chemical physics physical chemistry and related fields Combines a big picture overview of chemistry as it relates to physics and computer science including detailed guidance on tackling chemistry problems from both theoretical and computational perspectives Treats nuclear dynamics and electronic structure on the same footing in discussions of the Born Oppenheimer approximation Includes examples of scientific programming in modern Fortran for problems related to the modeling of chemical reaction dynamics and the analysis of chemical bonding **Ebook:** Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed ,2000

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thermoelectric properties of cluster based materials and clusters for CO2 activation and conversion before concluding with an incisive discussion of trends and directions likely to dominate the subject of superatoms in the coming years Readers will also benefit from the inclusion of A thorough introduction to the rational design of superatoms using electron counting rules Explorations of superhalogens endohedrally doped superatoms and assemblies and magnetic superatoms A practical discussion of atomically precise synthesis of chemically modified superatoms A concise treatment of superatoms as the building blocks of 2D materials as well as superatom based ferroelectrics and cluster based materials for energy harvesting and storage Perfect for academic researchers and industrial scientists working in cluster science energy materials thermoelectrics 2D materials and CO2 conversion Superatoms Principles Synthesis and Applications will also earn a place in the libraries of interested professionals in chemistry physics materials science and nanoscience **Harcourt Science:** Physical science, [grade] 5, Units E and F, teacher's ed ,2000 Misconceptions in Chemistry Hans-Dieter Barke, Al Hazari, Sileshi Yitbarek, 2008-11-18 Over the last decades several researchers discovered that children pupils and even young adults develop their own understanding of how nature really works These pre concepts concerning combustion gases or conservation of mass are brought into lectures and teachers have to diagnose and to reflect on them for better instruction In addition there are school made misconceptions concerning equilibrium acid base or redox reactions which originate from inappropriate curriculum and instruction materials. The primary goal of this monograph is to help teachers at universities colleges and schools to diagnose and cure the pre concepts In case of the school made misconceptions it will help to prevent them from the very beginning through reflective teaching The volume includes detailed descriptions of class room experiments and structural models to cure and to prevent these misconceptions Glencoe Physical Science McGraw Hill,1999 2000 2005 State Textbook Adoption Addison-Wesley Science Insights ,1996 **Prentice Hall Physical** Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Ionic Liquids in Chemical Analysis** Mihkel Koel, 2008-10-09 An Overview of a Rapidly Expanding Area in Chemistry Exploring the future in chemical analysis research Ionic Liquids in Chemical Analysis focuses on materials that promise entirely new ways to perform solution chemistry. It provides a broad overview of the applications of ionic liquids in various areas of analytical chemistry in **Targeted Radionuclide Therapy** Tod W. Speer, 2012-03-28 Radioimmunotherapy also known as systemic targeted radiation therapy uses antibodies antibody fragments or compounds as carriers to guide radiation to the targets It is a topic rapidly increasing in importance and success in treatment of cancer patients This book represents a comprehensive amalgamation of the radiation physics

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