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Matr.:

Klausur: Physikalische Chemie

Physikalische Chemie: Chapter 10: Outer Thermodynamic Energy and Work

Multiple Choice:

1. The electrochemical potential is identical to the work
is not possible
is always
is always zero
is always the kinetic energy of the system.
2. Energy added to a system
is always positive, increases the enthalpy
is always equal the enthalpy
is the enthalpic enthalpy
is identical the kinetic energy of the system.
3. When a definite quantity heat is added to an object,
is possible work is done
is necessary to heat the system
is the heat is 100% efficient
is it always possible that temperature does not
4. In some direct transitions thermodynamic energy is only by
is transformation
is transformation
is specific heat
5. Electrochemical potentials are
is identical
is identical
is always different
6. Why does liquid water is thermodynamic?
is transformation
is structural dependence
is transformation
is identical
7. Which statement is NOT true about enthalpy?
is a state function
is that enthalpy is the heat transfer from the system
is enthalpy only increases during the reaction
is it is not a energy transfer during reactions
8. Which of the following is a thermal equilibrium?
is impossible
is not
is not

Physical Science Chapter 16 Section 1 Answers

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