

Periodic Table Scavenger Hunt

Use your periodic table to help answer each question.

1. What do we mean by a "group" of elements?

On the periodic table, groups (or families) are the columns (going up and down). Elements in the same group have the same number of valence electrons and because of this, they behave the same way in chemical reactions.

2. Name the first and last element in group VIA (6A) or 16. Oxygen and Polonium

3. What do we mean by a "period" of elements?

On the periodic table, periods are the rows (going left to right). Elements in the same period have the same number of shells, but are not similar in properties.

4. Name the first and last element in period 3. Sodium and Argon

5. What is special about the group of noble gases?

Noble gases have a complete outer shell (maximum number of valence electrons in their valence shell). Because of this, they do not give, take, or share electrons with other atoms. They are very unreactive.

6. Where are the metals found in the periodic table? Left side of the staircase,

7. Where are the non-metals found in the periodic table? Right side of the staircase,

8. Which elements are found around the "staircase" of the periodic table?

Metalloids (Boron, Silicone, Germanium, Arsenic, Antimony, Tellurium)

9. Why are these elements at the "staircase" special?

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