The Periodic Table of Elements

In 1869 Dimitri Mendeleev organized the elements in a table based on the properties of the elements.



have the same number of valence electrons.

	id salment or	2					-	Course Sin
1 st Period → ⊙	1 H	Gerup 2.4	Group 31	Group 43	Green 54	Garate	b 000 76	2 He
	Hydrogen	ACCORDING TO	Allerton Co.	g inches(s)	PROPERTY.	ACAD SALES	The Paris	Helium
2nd Period → (i)	3 Li	Be	5 B	e C	M	8	9 F	10 Ne
	Lithium	beryllium	Boron	Carbon	Nitrogen	Oxygen	Flouring	Neon
300 Period - (1)	" Na	12 Ma	13 AI	14 Si	15 P	16 S	17 CI	18 Ar
(3 shells)	Sodium	Magnesiam	Aluminom	Silicon	Phesphores	Swifer	Chlorine	Argon
4th Period - (6)	19 K	20 Ca	31 Ga	32 Ge	As As	34 Se	35 Br	36 Kr
(4 Shells)	Potassium	Calcium	Gallium	Germanium				Krypton

{Periods} (rows)

The elements in each period have the same 3 number of shells.

Today, 114 elements are arranged by atomic number in the periodic table. The atomic number is based on the number of in an element.

Periodic Table Notes For Middle School

TD Snyder

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