

Name: _____

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CHEMISTRY

Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

The image shows a standard periodic table of elements. The elements are color-coded by groups:

- Group 1 (Alkali metals): Light blue
- Group 2 (Alkaline earth metals): Orange
- Groups 3-10 (Transition metals): Yellow
- Group 11 (Coinage metals): Light green
- Group 12 (Zn, Cd, Hg): Light blue
- Groups 13-18 (Main group elements): Various shades of green, yellow, and orange
- Lanthanides and Actinides: Light blue

Labels for Boron (B) and its isotopes are present:

- B**: Element symbol for Boron, located in Group 13, Period 2.
- Boron**: Element name, located below the symbol B.
- Atomic number**: 5, located to the left of the symbol B.
- Atomic mass**: 10.811, located to the right of the symbol B.
- Isotopes**: A label pointing to the isotopes of Boron, located to the right of the symbol B.
- Group**: A label pointing to the group of Boron, located to the right of the symbol B.
- Period**: A label pointing to the period of Boron, located to the right of the symbol B.

Isotopes of Boron are shown as small boxes with their mass numbers and natural abundances:

- ¹⁰B (19.9%)
- ¹¹B (80.1%)

Part 1: Atomic Number. On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- | | | | | | |
|----|--|-----------|-----|--|--------------|
| 1. | | Hydrogen | 7. | | Cerium |
| 2. | | Oxygen | 8. | | Iridium |
| 3. | | Magnesium | 9. | | Bismuth |
| 4. | | Vanadium | 10. | | Uranium |
| 5. | | Krypton | 11. | | Fermium |
| 6. | | Ruthenium | 12. | | Darmstadtium |

Periodic Table Answers Section 3

SA Adler



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