
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, she placed that question's correct statement or fact answer's letter in parentheses.

1. In developing his periodic table, Mendeleev's observations were
 - a. periodic elements.
 - b. elements in liquid form.
 - c. isotopes.
 - d. **properties.**
2. Mendeleev's periodic table showed all elements in order of
 - a. increasing atomic mass because he grouped together elements with similar
 - b. **properties.**
 - c. atomic numbers.
 - d. isotopes.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable isotopes.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number was
 - b. **discovered.**
 - c. Mendeleev.
 - d. discovered.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. isotopes.
 - c. ions.
 - d. elements.

Periodic Law Section Quiz Answers

Lei Shi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Shi'.

Periodic Law Section Quiz Answers:

Periodic Law Section Quiz Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Periodic Law Section Quiz Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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