

Quiz & Worksheet - Atomic and Ionic Radii on the Periodic Table

1. As you move from left to right on the periodic table, why does the atomic radius decrease?

- ☐ As more protons are added, the more tightly the surrounding electrons are drawn inwards.
- ☐ As more neutrons are added, the more tightly the surrounding electrons are drawn inwards.
- ☐ As more electrons are added, the more tightly they draw themselves towards the nucleus.
- ☐ As the mass of the nucleus rises, the more tightly the surrounding electrons are drawn inwards.

2. As you move from top to bottom on the periodic table, what happens to the atomic radius?

- ☐ It decreases.
- ☐ It decreases, then increases.
- ☐ It remains constant.
- ☐ It increases, then decreases.
- ☐ It increases.

3. Which of these elements has the largest atomic radius: argon, phosphorus, aluminum, sodium or magnesium?

- ☐ Argon (Ar)
- ☐ Phosphorus (P)
- ☐ Sodium (Na)
- ☐ Aluminum (Al)
- ☐ Magnesium (Mg)

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Periodic Elements Need To Knows Answer Key

Joacim Rocklöv



Periodic Elements Need To Knows Answer Key:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Periodic Elements Need To Knows Answer Key**

In a global inundated with screens and the cacophony of instant transmission, the profound energy and psychological resonance of verbal art often disappear in to obscurity, eclipsed by the constant onslaught of sound and distractions. However, nestled within the musical pages of **Periodic Elements Need To Knows Answer Key**, a interesting work of literary splendor that impulses with natural emotions, lies an remarkable trip waiting to be embarked upon. Composed with a virtuoso wordsmith, that interesting opus books visitors on an emotional odyssey, lightly revealing the latent possible and profound impact stuck within the delicate web of language. Within the heart-wrenching expanse with this evocative evaluation, we will embark upon an introspective exploration of the book is central styles, dissect their fascinating writing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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