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Exceed to Succeed

Worksheet: Periodic Trends

1. ATOMIC RADIUS

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li. Na. K
- c. Ge. P. O.
- d. C. N. Al
- e. Al. Cl. Ga.

IONIC RADIUS 2...

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a. Mg²⁺, Si⁴⁻, S²⁻ b. Mg²⁺, Ca²⁺, Ba²⁺
- c. F. Cl. Br
- d. Ba²⁺, Cu²⁺, Zn²⁺ e. Si⁴⁻, P³⁻, O²⁻

IONIZATION ENERGY 3...

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F. Cl. Br
- d. Ba. Cu. Ne
- e. Si. P. He

44. ELECTRONEGATIVITY

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C. O. Ne
- c. Si. P. O
- d. K. Mg, P
- e. S. F. He

<u>Periodic Trends Worksheet 1 Using The Periodic Law</u> <u>Answers</u>

Wolfgang Guggemos

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In some sort of inundated with monitors and the cacophony of instantaneous connection, the profound energy and mental resonance of verbal artistry often disappear in to obscurity, eclipsed by the constant onslaught of sound and distractions. Yet, situated within the musical pages of **Periodic Trends Worksheet 1 Using The Periodic Law Answers**, a charming perform of fictional splendor that impulses with fresh feelings, lies an memorable trip waiting to be embarked upon. Written by a virtuoso wordsmith, this interesting opus books readers on a mental odyssey, lightly exposing the latent potential and profound influence stuck within the elaborate web of language. Within the heart-wrenching expanse of the evocative evaluation, we can embark upon an introspective exploration of the book is key subjects, dissect its interesting writing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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