

Periodic Trends Worksheet

Name: _____ Grade: _____ ID: _____

Directions: Use your notes to answer the following questions.

- 1 Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Ans: O, C, Al, K

- 2 Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Ans: Ne, Al, S, O

- 3 Why does fluorine have a higher ionization energy than iodine?

Ans: Fluorine has a smaller atomic radius, so the proton can exert a greater pull on electron (Valence shell is close)

- 4 Why do elements in the same family generally have similar properties?

Ans: They have the same valence shell electron arrangement.

- 5 Indicate whether the following properties increase or decrease from left to right across the periodic table.

- a. atomic radius (excluding noble gases) **Decrease**
- b. first ionization energy **Increase**
- c. electronegativity **Increase**

- 6 What trend in atomic radius occurs across the periodic table? What causes this trend?

Ans: Atomic radius increases down a group because energy levels (shells) are added.

- 7 What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ans: Ionization energy increases across a period because as elements become less metallic, it requires more energy to remove an electron.

- 8 Circle the atom in each pair that has the largest radius.

a. Al or B

d. O or F

b. Na or Al

e. Br or Cl

c. S or O

f. Mg or Ca

Periodic Table Trends Answer Key

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Constructing Measures Mark Wilson, 2004-12-13 Constructing Measures introduces a way to understand the advantages and disadvantages of measurement instruments how to use such instruments and how to apply these methods to develop new instruments or adapt old ones The book is organized around the steps taken while constructing an instrument It opens with a summary of the constructive steps involved Each step is then expanded on in the next four chapters These chapters develop the building blocks that make up an instrument the construct map the design plan for the items the outcome space and the statistical measurement model The next three chapters focus on quality control They rely heavily on the calibrated construct map and review how to check if scores are operating consistently and how to evaluate the reliability and validity evidence The book introduces a variety of item formats including multiple choice open ended and performance items projects portfolios Likert and Guttman items behavioral observations and interview protocols Each chapter includes an overview of the key concepts related resources for further investigation and exercises and activities Some chapters feature appendices

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Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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Chemistry in a Month John Wilkinson, 2003 *Shriver and Atkins' Inorganic Chemistry* Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student s chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate the dynamic nature of inorganic chemistry

Build the Brain the Common Core Way Pamela Nevills, 2014-04-03 Brain compatible learning techniques to help you motivate and challenge your students Boldly take hold of the new Common Core expectations with this fresh innovative resource full of practical tips from international educational expert Pamela Nevills Find out what's new what's expected and how understanding the connection between neuroscience and deep learning is the key to Common Core success Nevills provides cutting edge easy to implement ideas that beg to be used immediately Learn to Skillfully integrate the standards with current educational practices Understand the science behind Common Core requirements Plan innovative and creative activities to help students learn about learning Master teacher lead student centered 21st Century Learning activities Includes school vignettes sample lessons K 12 links and vivid brain imagery that diagram how deep learning happens Revolutionize your teaching and prepare students for success in school and beyond with this encouraging easy to read guide Build the Brain the Common Core Way is an outstanding resource for teachers who are implementing our new standards It provides practical examples that demonstrate how to turn the classroom into a student centered learning environment where high expectations abound fostering higher levels of student achievement Melissa Miller Science History Instructor Randall G Lynch Middle School Farmington AR There are examples not only of ways to integrate brain based pedagogy and the Common Core Standards but also examples from across disciplines The author doesn't just focus on one area of expertise but integrates core and elective curriculum into the fabric of each chapter Deanna Brunlinger Science Teacher Elkhorn Area High School WI

Inorganic Chemistry Mark Weller, Mark T. Weller, Tina Overton, Jonathan Rourke, Fraser Armstrong, 2014 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Sixth Edition is the ideal course companion for the duration of a student's degree The authors have drawn upon their extensive teaching and research experience in updating this established text the sixth edition retains the much praised clarity of style and layout from previous editions while offering an enhanced Frontiers section Exciting new applications of inorganic chemistry have been added to this section in particular relating to materials chemistry and medicine This edition also sees a greater use of learning features to provide students with all the support they need for their studies Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resource Centre For registered adopters of the text Figures marginal structures and tables of data ready to download Test bank For students Answers to self tests and exercises from the book Videos of chemical reactions Tables for group theory Web links Interactive structures and other resources on www.chemtube3d.com

Chemistry at a Glance Roger Owen, Sue King, 2005-06-27 This book aims to cover the specifications of the main examination boards for GCSE Double Science GCSE Single Science and the core content of GCSE Chemistry Where relevant Key Stage 3 material is summarized as an introduction to GCSE topics This serves as revision of work done prior to Key Stage 4 and a foundation for GCSE studies The book is [NEET Chemistry 1500+ MCQs](#) Disha

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