

Name: _____ Date: _____

Periodic Table Scavenger Hunt

Search your periodic table for the answers to these questions.

- | | |
|--------------------|--|
| <u>Silicon</u> | 1. Which element is number 14 on the periodic table? |
| <u>Cf</u> | 2. What is the element symbol for californium? |
| <u>83</u> | 3. How many protons are in an atom of bismuth? |
| <u>Noble Gas</u> | 4. To which element group does argon belong? |
| <u>Cadmium</u> | 5. Which element would you expect to have a higher mass: cadmium or zinc? |
| <u>12.01</u> | 6. What is the atomic mass of carbon? |
| <u>Lanthanides</u> | 7. What do you call the element series from atomic number 57-71? |
| <u>Gold</u> | 8. Which element has a symbol that starts with a letter different from the first one in its name: aluminum, copper, gold, rhenium? |
| <u>Hydrogen</u> | 9. Which element has the lowest atomic mass? |
| <u>Ruthenium</u> | 10. What is the first element with an atomic mass greater than 100? |
| <u>Aluminum</u> | 11. What is the first basic metal on the periodic table? |
| <u>False</u> | 12. True or false: Tin and antimony are in the same element group. |
| <u>Francium</u> | 13. What is the heaviest alkali metal? |
| <u>12</u> | 14. How many protons are in an atom of magnesium? |
| <u>Silicon</u> | 15. Which of the following is not a nonmetal: sulfur, oxygen, silicon, nitrogen? |
| <u>Tungsten</u> | 16. What is the name of the element with the symbol W? |
| <u>Palladium</u> | 17. Which element has an atomic mass of 106.42? |
| <u>Halogen</u> | 18. Astatine belongs to which element group: nonmetal, halogen, noble gas? |
| <u>Barium</u> | 19. What is the element with the symbol Ba? |
| <u>I or O</u> | 20. Name a letter never used in any element symbol? |

Periodic Table Scavenger Hunt Worksheet With Answers

Linda Trombley

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Griffin,2018-12-15 The periodic table is the way scientists have organized the known elements but up and coming scientists can learn much from the periodic table as well Your readers will learn how to read the periodic table including what an element s atomic number means and what the rows and columns signify They ll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table Fact boxes and images add additional information to the accessible text Mastering the Periodic Table Linda Trombley,2000

The Basics of the Periodic Table Leon Gray,2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table Included are discussions of the discovery of matter atoms atomic structure molecules compounds ions and isotopes as well as the first identifications of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events Periodic Table Cindy Blobaum,2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket *Mystery of the Periodic Table* Benjamin D Wiker,2003-04-18 Leads the reader on a delightful and absorbing journey through the ages on the trail of the elements of the Periodic Table as we know them today

He introduces the young reader to people like Von Helmont Boyle Stahl Priestly Cavendish Lavoisier and many others all incredibly diverse in personality and approach who have laid the groundwork for a search that is still unfolding to this day The first part of Wiker s witty and solidly instructive presentation is most suitable to middle school age while the later chapters are designed for ages 12 13 and up with a final chapter somewhat more advanced Illustrated by Jeanne Bendick and Ted Schluenderfritz **The Periodic Table** Adrian Dingle,2011 Designed to make learning chemistry much easier and a whole lot more fun these elements show you the periodic table as you have never seen it before Every element in this engaging little book is a specially created character with its own unique personality *Atoms, Molecules & Elements Gr. 5-8* George Graybill,2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included **The Periodic Table** Marie Roesser,2025-07-30 Understanding the elements gets a whole lot easier when you understand the periodic table For years scientists tried to make sense of the chemical world through a visually accurate representation Today s table is organized by increasing atomic number or the number of protons in the atomic nucleus This chart has aided in the discovery of new elements the most recent is oganesson with 118 protons Through this valuable volume budding scientists will learn their way around the periodic table including what an element s atomic mass is and what the rows and columns signify Fact boxes and labeled images further illuminate difficult curricular concepts Understanding the Periodic Table Jane P. Gardner,2018 Introduces the periodic table of the elements

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