

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
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Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data						2	
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd						He	
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																											
speed of light in vacuum c = 299 792 458 m s ⁻¹ (exact)																											
Planck constant h = 6.626 070 15 × 10 ⁻³⁴ J s (exact)																											
elementary charge e = 1.602 176 634 × 10 ⁻¹⁹ C																											
electron mass m _e = 9.109 383 56 × 10 ⁻³¹ kg																											
proton mass m _p = 1.672 621 9 × 10 ⁻²⁷ kg																											
fine-structure constant α = 1/137.036																											
Rydberg constant R _∞ = 10 973 731.568 162 m ⁻¹																											
Bohrmann constant k _B = 1.380 658 × 10 ⁻²³ J K ⁻¹																											

Atomic Number: 58
Ground-state Level: $4f^7 5d^1$
Symbol: **Ce**
Name: Cerium
Atomic Weight: 140.116
Ground-state Configuration: $[Xe]4f^7 5d^1$
Ionization Energy (eV): 5.5387

¹Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

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LP Steffe



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








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