

# The Periodic Law

## SECTION 2

Select the correct answer to the following questions for the space provided.

1. \_\_\_\_\_ for the modern periodic table, elements are ordered
  - (a) according to increasing atomic mass
  - (b) according to Mendeleev's original design
  - (c) according to increasing atomic number
  - (d) based on how they were discovered
2. \_\_\_\_\_ Mendeleev's original short periodic table gave useful chemical properties of elements appeared in regular intervals and elements discovered later completed intervals of his design.
  - (a) slowly
  - (b) not long
  - (c) almost instantly
  - (d) almost never
3. \_\_\_\_\_ The modern periodic law states that
  - (a) all new elements with the same properties are listed in the same group every year
  - (b) the physical and chemical properties of elements are functions of their atomic numbers
  - (c) elements exhibit properties of both periods and groups
  - (d) the chemical properties of elements can be grouped according to periodicity but physical properties cannot
4. \_\_\_\_\_ The discovery of the noble gases changed Mendeleev's periodic table.
  - (a) adding a row
  - (b) adding a column
  - (c) adding a new group
  - (d) adding a new period
5. \_\_\_\_\_ The noble gases that property of the noble gases is that they are
  - (a) unstable
  - (b) inert
  - (c) radioactive
  - (d) highly reactive
6. \_\_\_\_\_ Likewise, the first column in Group 1 has characteristic elements of 1. The second column in the group has characteristic elements of
  - (a) 2
  - (b) 10
  - (c) 18
  - (d) 32
7. How groups in Mendeleev's periodic table of 18 and the elements of 18
  - (a) 18 How many elements are in this group?
  - (b) 18 How many elements are in this group?
  - (c) 18 How many elements are in this group?
  - (d) 18 How many elements are in this group?

# Periodic Law Chapter 5 Test Answers

**Brendan G. Carr**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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