

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-4)

Remember: Limiting reagent is the reactant that will completely react and leave another reactant

Excess: All reactants except the limiting reagent will be left over. Remember, limiting reagent is the reactant that will completely react.

Remember, limiting reagent will determine how much product is produced and byproducts will be the leftover reactant in a chemical reaction.



The ratio required for the components for the chemical reaction.

2H ₂ (g)	100.00 grams
O ₂ (g)	50.00 grams
2H ₂ O(l)	100.00 grams
Excess	100.00 grams

1. 100.00 g of each component will be used & react to form 100.00 g of product water. 100.00 g of hydrogen & 50.00 g of oxygen will be used. 100.00 g of water is the product.



2. 100.00 g of each component will be used & react to form 100.00 g of product water. 100.00 g of hydrogen & 50.00 g of oxygen will be used. 100.00 g of water is the product.

3. 100.00 g of each component will be used & react to form 100.00 g of product water. 100.00 g of hydrogen & 50.00 g of oxygen will be used. 100.00 g of water is the product.



4. 100.00 g of each component will be used & react to form 100.00 g of product water. 100.00 g of hydrogen & 50.00 g of oxygen will be used. 100.00 g of water is the product.



Percent Yield And Limiting Reactants Quiz

TD Snyder



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