

SECOND MID TERM TEST - 2024

Standard XII

Reg.No.

01256

CHEMISTRY

Time : 1.30 hrs

Part - I

Marks :

10 x 1 =

1. Choose the correct answer:

- A magnetic moment of 1.73 BM will be shown by one among the following
a) TiCl_4 b) $[\text{CoCl}_6]^{4-}$ c) $[\text{Cu}(\text{NH}_3)_4]^{2+}$ d) $[\text{Ni}(\text{CN})_4]^{2-}$
- Which type of isomerism is exhibited by $[\text{Pt}(\text{NH}_3)_2 \text{Cl}_2]$?
a) coordination isomerism b) linkage isomerism
c) optical isomerism d) geometrical isomerism
- The number of electrons that have a total charge of 9650 coulombs is
a) 6.22×10^{23} b) 6.022×10^{24} c) 6.022×10^{22} d) 6.022×10^{-34}
- Which of the following electrolytic solution has the least specific conductance?
a) 2 N b) 0.002 N c) 0.02 N d) 0.2 N

Physical Chemistry Second Midtterm Exam Questions

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Table of Contents Physical Chemistry Second Midtterm Exam Questions

1. Understanding the eBook Physical Chemistry Second Midtterm Exam Questions
 - The Rise of Digital Reading Physical Chemistry Second Midtterm Exam Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Chemistry Second Midtterm Exam Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Chemistry Second Midtterm Exam Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Chemistry Second Midtterm Exam Questions
 - Personalized Recommendations
 - Physical Chemistry Second Midtterm Exam Questions User Reviews and Ratings
 - Physical Chemistry Second Midtterm Exam Questions and Bestseller Lists

5. Accessing Physical Chemistry Second Midtterm Exam Questions Free and Paid eBooks
 - Physical Chemistry Second Midtterm Exam Questions Public Domain eBooks
 - Physical Chemistry Second Midtterm Exam Questions eBook Subscription Services
 - Physical Chemistry Second Midtterm Exam Questions Budget-Friendly Options
6. Navigating Physical Chemistry Second Midtterm Exam Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Chemistry Second Midtterm Exam Questions Compatibility with Devices
 - Physical Chemistry Second Midtterm Exam Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Chemistry Second Midtterm Exam Questions
 - Highlighting and Note-Taking Physical Chemistry Second Midtterm Exam Questions
 - Interactive Elements Physical Chemistry Second Midtterm Exam Questions
8. Staying Engaged with Physical Chemistry Second Midtterm Exam Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Chemistry Second Midtterm Exam Questions
9. Balancing eBooks and Physical Books Physical Chemistry Second Midtterm Exam Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Chemistry Second Midtterm Exam Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Chemistry Second Midtterm Exam Questions
 - Setting Reading Goals Physical Chemistry Second Midtterm Exam Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Chemistry Second Midtterm Exam Questions
 - Fact-Checking eBook Content of Physical Chemistry Second Midtterm Exam Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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