

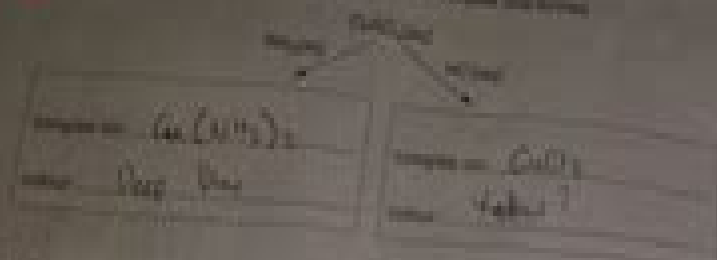
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the synthesis of a copper complex.

In the boxes, write the formula and name of the ligands and solvent.



(b) Cu^{2+} has two d-orbitals in the complex with two unpaired electrons and two water molecules. The d-orbitals are in a tetrahedral field.

The ground state of the d-orbitals in A is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term tetrahedral field?

A tetrahedral field is a field where the central metal ion is surrounded by four ligands in a tetrahedral arrangement. The d-orbitals are split into two sets of two orbitals each, with the higher energy set containing two unpaired electrons.

(d) The complex is a copper(II) complex.

The reaction is about the synthesis of a copper complex. The reaction is shown below.



Complex A

Complete the d-orbital diagram of the other two complexes of A. You do not need to include any charges.

Indicate with ticks whether the complexes are oct, tet, square, tetrahedral or a combination of these types.

Structure													
Type <table border="1"> <tr> <td>oct</td> <td></td> </tr> <tr> <td>sq.pl</td> <td>☒</td> </tr> <tr> <td>tetrahedral</td> <td></td> </tr> </table>	oct		sq.pl	☒	tetrahedral		<table border="1"> <tr> <td>oct</td> <td>☒</td> </tr> <tr> <td>sq.pl</td> <td></td> </tr> <tr> <td>tetrahedral</td> <td>☒</td> </tr> </table>	oct	☒	sq.pl		tetrahedral	☒
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(e) What is the empirical formula, including the charge, of the complex in A?



Ocr Chemistry F32may 2013 Unofficial

Thomas Griffiths



Ocr Chemistry F32may 2013 Unofficial:

The Enigmatic Realm of **Ocr Chemistry F32may 2013 Unofficial**: Unleashing the Language is Inner Magic

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Table of Contents **Ocr Chemistry F32may 2013 Unofficial**

1. Understanding the eBook **Ocr Chemistry F32may 2013 Unofficial**
 - The Rise of Digital Reading **Ocr Chemistry F32may 2013 Unofficial**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Ocr Chemistry F32may 2013 Unofficial**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Ocr Chemistry F32may 2013 Unofficial**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Ocr Chemistry F32may 2013 Unofficial**
 - Personalized Recommendations
 - **Ocr Chemistry F32may 2013 Unofficial** User Reviews and Ratings
 - **Ocr Chemistry F32may 2013 Unofficial** and Bestseller Lists

5. Accessing Ocr Chemistry F32may 2013 Unofficial Free and Paid eBooks
 - Ocr Chemistry F32may 2013 Unofficial Public Domain eBooks
 - Ocr Chemistry F32may 2013 Unofficial eBook Subscription Services
 - Ocr Chemistry F32may 2013 Unofficial Budget-Friendly Options
6. Navigating Ocr Chemistry F32may 2013 Unofficial eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr Chemistry F32may 2013 Unofficial Compatibility with Devices
 - Ocr Chemistry F32may 2013 Unofficial Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry F32may 2013 Unofficial
 - Highlighting and Note-Taking Ocr Chemistry F32may 2013 Unofficial
 - Interactive Elements Ocr Chemistry F32may 2013 Unofficial
8. Staying Engaged with Ocr Chemistry F32may 2013 Unofficial
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Chemistry F32may 2013 Unofficial
9. Balancing eBooks and Physical Books Ocr Chemistry F32may 2013 Unofficial
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Chemistry F32may 2013 Unofficial
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr Chemistry F32may 2013 Unofficial
 - Setting Reading Goals Ocr Chemistry F32may 2013 Unofficial
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Chemistry F32may 2013 Unofficial
 - Fact-Checking eBook Content of Ocr Chemistry F32may 2013 Unofficial
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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