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## OCR A LEVEL CHEMISRTY EXAM 2019 WITH 100% CORRECT QUESTIONS AND ANSWERS

**alkali - ANSWER-**A type of base that dissolves in water forming hydroxide ions, OH<sup>-</sup> (aq) ions.

**alkanes - ANSWER-**The homologous series with the general formula: C<sub>n</sub>H<sub>2n+2</sub>.

**alkyl group - ANSWER-**An alkane with a hydrogen atom removed, e.g. CH<sub>3</sub>, C<sub>2</sub>H<sub>5</sub>; alkyl groups are often shown as 'R'.

**amount of substance - ANSWER-**The quantity whose unit is the mole. Chemists use 'amount of substance' as a means of counting atoms.

**anhydrous - ANSWER-**A substance that contains no water molecules.

**anion - ANSWER-**A negatively charged ion.

**atom economy - ANSWER-**atom economy =  $\frac{\text{molecular mass of the desired product}}{\text{sum of molecular masses of all products}} \times 100$

**atomic orbital - ANSWER-**A region within an atom that can hold up to two electrons, with opposite spins.

**atomic (proton) number - ANSWER-**The number of protons in the nucleus of an atom

**acid - ANSWER-**A species that is a proton donor.

**activation energy - ANSWER-**The minimum amount of kinetic energy that particles need to have in order to react when they collide

**addition polymer - ANSWER-**A very long molecular chain formed by repeated addition reactions of many unsaturated alkene molecules (monomers).

**addition polymerisation - ANSWER-**The process in which unsaturated alkene molecules (monomers) add on to a growing polymer chain one at a time to form a very long saturated molecular chain (the addition polymer).

**addition reaction - ANSWER-**A reaction in which a reactant is added to an unsaturated molecule to make a saturated molecule.

# Ocr Chemisrty Gcse Higher May 2013

**Richard Parsons**



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