

A2 Unit F334: Chemistry of Materials

Topic 3: What's in a medicine? (date)

A study of medicines such as aspirin, their development, chemistry and synthesis, illustrating some of the features of the pharmaceutical industry. The chemical ideas in this module are:

- Homolytic and heterolytic acids, bases, coloured compounds
- Acid-base reactions
- Medicines: manufacture and testing
- IR spectroscopy and mass spectroscopy

Topic	Alt code	Number	Assessable learning outcomes	Number of questions	OSP A2 Revision guide (Page number)	Chemical changes (Page number)	Chemical ideas (Page number)
Equilibria		10401	Describe acids in terms of the Brønsted-Lowry theory as proton donors, and bases as proton acceptors	1	6		100
		10402	Identify the proton donor and proton acceptor in an acid-base reaction	1	6		100
Bonding and structure	11.1.0	10403	Draw and interpret simple electron-dot-and-cross diagrams to show how atoms (bond through one, covalent and three covalent bonds)	2	14		101-12
	11.1.0a	10404	Describe a simple model of metallic bonding		1		41
	12.1.0	10405	Use the electron-dot-repulsion principle to predict and explain the shapes of simple molecules (such as CH ₄ , NH ₃ , H ₂ O and HF) and ions (such as NH ₄ ⁺) with up to six outer pairs of electrons. (any combination of bonding pairs and lone pairs) (no treatment of hybridisation or molecular orbitals is expected)	1	7		101-105

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Table of Contents Ocr B F334 January 2009 Mark Scheme

1. Understanding the eBook Ocr B F334 January 2009 Mark Scheme
 - The Rise of Digital Reading Ocr B F334 January 2009 Mark Scheme
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr B F334 January 2009 Mark Scheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr B F334 January 2009 Mark Scheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr B F334 January 2009 Mark Scheme

- Personalized Recommendations
- Ocr B F334 January 2009 Mark Scheme User Reviews and Ratings
- Ocr B F334 January 2009 Mark Scheme and Bestseller Lists
- 5. Accessing Ocr B F334 January 2009 Mark Scheme Free and Paid eBooks
 - Ocr B F334 January 2009 Mark Scheme Public Domain eBooks
 - Ocr B F334 January 2009 Mark Scheme eBook Subscription Services
 - Ocr B F334 January 2009 Mark Scheme Budget-Friendly Options
- 6. Navigating Ocr B F334 January 2009 Mark Scheme eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr B F334 January 2009 Mark Scheme Compatibility with Devices
 - Ocr B F334 January 2009 Mark Scheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr B F334 January 2009 Mark Scheme
 - Highlighting and Note-Taking Ocr B F334 January 2009 Mark Scheme
 - Interactive Elements Ocr B F334 January 2009 Mark Scheme
- 8. Staying Engaged with Ocr B F334 January 2009 Mark Scheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr B F334 January 2009 Mark Scheme
- 9. Balancing eBooks and Physical Books Ocr B F334 January 2009 Mark Scheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr B F334 January 2009 Mark Scheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr B F334 January 2009 Mark Scheme
 - Setting Reading Goals Ocr B F334 January 2009 Mark Scheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr B F334 January 2009 Mark Scheme

- Fact-Checking eBook Content of Ocr B F334 January 2009 Mark Scheme
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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