SAMPLE PAPER TEST 02 FOR BOARD EXAM 2023

SUBJECT: SCIENCE CLASS: X

MAX. MARKS: 80 DURATION: 3 HRS

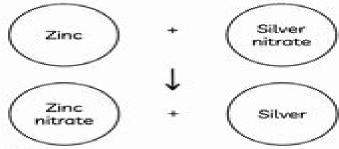
General Instruction:

- 1. This Question Paper has 5 Sections A-E.
- 2. Section A has 20 MCQs carrying 1 mark each.
- Section B has 5 questions carrying 02 marks each.
- Section C has 6 questions carrying 03 marks each.
- 5. Section D has 4 questions carrying 05 marks each.
- 6. Section E has 3 case based integrated units of assessment (04 marks each) with sub-parts of the values of 1, 1 and 2 marks each respectively.
- 7. All Questions are compulsory. However, an internal choice in 2 Qs of 5 marks, 2 Qs of 3 marks and 2 Questions of 2 marks has been provided. An internal choice has been provided in the 2marks questions of Section E
- 8. Draw neat figures wherever required. Take $\pi = 22/7$ wherever required if not stated.

SECTION - A

Questions 1 to 20 carry 1 mark each.

1. Which of the following a balanced equation for the following chemical reaction:



- (a) $Zn + Ag(NO_3)_2 \rightarrow Zn(NO_3)_2 + 2Ag$
- (b) $2Zn + 2AgNO_3 \rightarrow Zn(NO_3)_2 + 2Ag$
- (c) $Zn + 2AgNO_3 \rightarrow Zn(NO_3)_2 + 2Ag$
- (d) $Zn + AgNO_3 \rightarrow Zn(NO_3)_2 + Ag$
- 2. Which of the following will turn phenolphthalein pink?
 - (a) NaOH(aq) (b) HCl(aq) (c) CH3COOH(aq) (d) H2O
- 3. While studying the saponification reaction, what do you observe when you mix an equal amount of colourless vegetable oil and 20% aqueous solution of NaOH in a beaker?
 - (a) The colour of the mixture has become dark brown.
 - (b) A brisk effervescence is taking place in the beaker.
 - (c) The outer surface of the beaker has become hot.
 - (d) The outer surface of the beaker has become cold.
- 4. Why do we store silver chloride in dark-coloured bottles?
 - (a) To prevent precipitation of silver chloride
 - (b) To prevent decomposition of silver chloride
 - (c) To promote decomposition of silver chloride
 - (d) All of these

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