



GRADE 12 PHYSICAL SCIENCES
NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

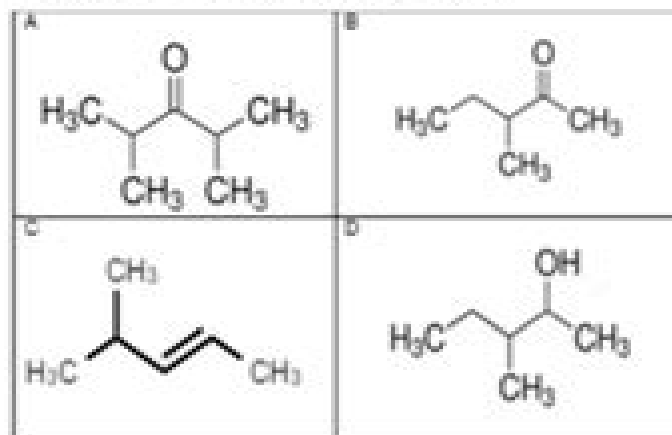
QUESTION 1

1.1 Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION NUMBER (1.1.1-1.1.10) on your answer sheet, for example 1.1.11. D.

1.1.1 Catalyst required for a thermal cracking:

- A Strong acids
- B No catalyst required
- C Pt, Ni or Pd
- D FCC

1.1.2 The correct molecular structure for 3-methyl-2-pentanone:



1.1.3 Suffix for the homologous series of esters:

- A -ene
- B -oate
- C -oic acid
- D -al

1.1.4 The relationship between volume and pressure of a gas is:

- A Constant
- B Directly proportional
- C Inversely proportional
- D Independent

1.1.5 Organic homologous series with a double bond between oxygen and carbon:

- A Alkenes
- B Aldehydes
- C Alkynes
- D Alcohols

1.1.6 Which of the following will cause a **DECREASE** in reaction rate?

- A Presence of a catalyst
- B Increase in the concentration of the reactant
- C Decrease in the concentration of the reactant
- D Alteration of electrical conductivity of reactant particles

1.1.7 The product concentration of a chemical reaction is decreased. What would the result be referring to Le Chatelier's principles.

- A The reaction rate of the forward reaction would increase.
- B The reaction rate of the backward reaction would increase.
- C The activation energy would decrease.
- D The activation energy would increase.

1.1.8 Which of the following scenarios would **NOT** cause a change in the Maxwell Boltzmann distribution of a gas?

- A Temperature changes.
- B Concentration changes.
- C Adding a catalyst.
- D Pressure changes.

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Chemistry

Thirumalaisamy P.
Velavancorresponding



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Table of Contents Physical Science Paper2 Caps Dbe November 2014 Chemistry

1. Understanding the eBook Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - The Rise of Digital Reading Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Personalized Recommendations
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry User Reviews and Ratings
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry and Bestseller Lists
5. Accessing Physical Science Paper2 Caps Dbe November 2014 Chemistry Free and Paid eBooks
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry Public Domain eBooks
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry eBook Subscription Services
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry Budget-Friendly Options
6. Navigating Physical Science Paper2 Caps Dbe November 2014 Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry Compatibility with Devices
 - Physical Science Paper2 Caps Dbe November 2014 Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Highlighting and Note-Taking Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Interactive Elements Physical Science Paper2 Caps Dbe November 2014 Chemistry
8. Staying Engaged with Physical Science Paper2 Caps Dbe November 2014 Chemistry

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Physical Science Paper2 Caps Dbe November 2014 Chemistry
- 9. Balancing eBooks and Physical Books Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Paper2 Caps Dbe November 2014 Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Setting Reading Goals Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Fact-Checking eBook Content of Physical Science Paper2 Caps Dbe November 2014 Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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 $\text{GaBr} + \text{NaBr} \rightarrow \text{SnO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{S}_3 + \text{NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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