

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and which from it? (anion/cation) _____
which cation is gained and electron is (anion/cation) _____
2. How placed in the halogens group has _____ valence electrons, and number is _____ 1 number in group 16. This is because of the _____ rules.
3. How _____ bond in the addition between oppositely charged ions.
ionic is called bond is _____
4. On a covalent bond, shared _____ electrons, referred to as
bond bond, all are _____ electrons.
5. 2-bonded group Which bond from the shared completely is a covalent bond is the _____, and which group is shared equally is the _____
How can predict if a compound is ionic, polar covalent, or nonpolar covalent by
looking at the _____
6. How many valence electrons for _____ valence electrons. This family contains all the
periodic table that all have 8 valence electrons is called _____
7. Shared pairs of electrons, or _____, by between two or
more _____

B. Formulas for Ionic Compounds - Put in the right order

Write the correct electron configuration of	Aluminum	Fluorine	Phosphorus	Lithium

3. Write in the chemical formula for each of the following ions:
anion/cation

Write your own ionic compounds, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.

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Table of Contents Physical Science Chemical Bonding Study Guide

1. Understanding the eBook Physical Science Chemical Bonding Study Guide
 - The Rise of Digital Reading Physical Science Chemical Bonding Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chemical Bonding Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chemical Bonding Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chemical Bonding Study Guide
 - Personalized Recommendations
 - Physical Science Chemical Bonding Study Guide User Reviews and Ratings
 - Physical Science Chemical Bonding Study Guide and Bestseller Lists
5. Accessing Physical Science Chemical Bonding Study Guide Free and Paid eBooks
 - Physical Science Chemical Bonding Study Guide Public Domain eBooks
 - Physical Science Chemical Bonding Study Guide eBook Subscription Services
 - Physical Science Chemical Bonding Study Guide Budget-Friendly Options
6. Navigating Physical Science Chemical Bonding Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Physical Science Chemical Bonding Study Guide Compatibility with Devices
- Physical Science Chemical Bonding Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chemical Bonding Study Guide
 - Highlighting and Note-Taking Physical Science Chemical Bonding Study Guide
 - Interactive Elements Physical Science Chemical Bonding Study Guide
- 8. Staying Engaged with Physical Science Chemical Bonding Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chemical Bonding Study Guide
- 9. Balancing eBooks and Physical Books Physical Science Chemical Bonding Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chemical Bonding Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Chemical Bonding Study Guide
 - Setting Reading Goals Physical Science Chemical Bonding Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Chemical Bonding Study Guide
 - Fact-Checking eBook Content of Physical Science Chemical Bonding Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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