

- (d) The student found that 22.50 cm^3 of $0.0800\text{ mol dm}^{-3}$ $\text{Na}_2\text{B}_4\text{O}_7$ reacted with 25.00 cm^3 of dilute hydrochloric acid.



- (i) Calculate the amount, in mol, of $\text{Na}_2\text{B}_4\text{O}_7$ used.

amount = mol [1]

- (ii) Calculate the amount, in mol, of HCl used.

amount = mol [1]

- (iii) Calculate the concentration, in mol dm^{-3} , of the HCl .

concentration = mol dm^{-3} [1]

[Total: 12]

END OF QUESTION PAPER

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Peter Cann, Peter Hughes



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Table of Contents Ocr Chemistry Jan 2013 Paper

1. Understanding the eBook Ocr Chemistry Jan 2013 Paper
 - The Rise of Digital Reading Ocr Chemistry Jan 2013 Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry Jan 2013 Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Chemistry Jan 2013 Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry Jan 2013 Paper
 - Personalized Recommendations
 - Ocr Chemistry Jan 2013 Paper User Reviews and Ratings

- Ocr Chemistry Jan 2013 Paper and Bestseller Lists
- 5. Accessing Ocr Chemistry Jan 2013 Paper Free and Paid eBooks
 - Ocr Chemistry Jan 2013 Paper Public Domain eBooks
 - Ocr Chemistry Jan 2013 Paper eBook Subscription Services
 - Ocr Chemistry Jan 2013 Paper Budget-Friendly Options
- 6. Navigating Ocr Chemistry Jan 2013 Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr Chemistry Jan 2013 Paper Compatibility with Devices
 - Ocr Chemistry Jan 2013 Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry Jan 2013 Paper
 - Highlighting and Note-Taking Ocr Chemistry Jan 2013 Paper
 - Interactive Elements Ocr Chemistry Jan 2013 Paper
- 8. Staying Engaged with Ocr Chemistry Jan 2013 Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Chemistry Jan 2013 Paper
- 9. Balancing eBooks and Physical Books Ocr Chemistry Jan 2013 Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Chemistry Jan 2013 Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr Chemistry Jan 2013 Paper
 - Setting Reading Goals Ocr Chemistry Jan 2013 Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr Chemistry Jan 2013 Paper
 - Fact-Checking eBook Content of Ocr Chemistry Jan 2013 Paper
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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