

Answer Key

Unit 1 Biochemistry

Answers to Unit 1 Preparation Questions Assessing Student Readiness (Student textbook pages 4–7)

- Sketches should depict a central atomic nucleus surrounded by four valence electrons (or six electrons in total), or six neutrons and six protons within the nucleus.
- b
- a
- metal, non-metal ion
- Since the atoms in a glucose molecule share valence electrons, the bonds between these atoms must be covalent.
 - Glucose is a stable molecule because its covalent bonds are stable and strong.
- water—molecular, atoms (both non-metals) are joined by covalent bonds
 - carbon dioxide—molecular, atoms (both non-metals) are joined by covalent bonds
 - glucose—molecular, atoms (two non-metals and uncharged hydrogen) are joined by covalent bonds
 - sodium chloride—ionic, ions (a metal and a non-metal) joined by an ionic bond
 - calcium phosphate—ionic, ions (a metal and a polyatomic ion) joined by ionic bonds
 - methane—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
 - oxygen—molecule or molecular element, atoms joined by double covalent bond, but not a molecular compound because it is formed from only one element (two oxygen atoms)
 - ammonia—molecular, atoms (a non-metal and uncharged hydrogen) joined by covalent bonds
- chloride
 - sulphate/sulfate
 - magnesium ion
 - copper (II) ion
- OH^-
 - S^{2-}
 - Al^{3+}
 - Fe^{3+}
- H_2O or water
- $\text{water} \rightarrow \text{hydrogen} + \text{oxygen}$
- d
- sodium hydroxide + hydrochloric acid $\rightarrow$
water + sodium chloride
- Since the pH of the solution increased, the student must have added a base to the solution.
 - One of: When the second substance was added, it released hydroxide ions (OH^-), which combined with protons (H^+) to produce water. The second substance accepted protons (H^+) from the solution.
- decomposition
 - neutralization
 - decomposition
 - synthesis
 - neutralization
 - synthesis
 - decomposition
 - synthesis
 - synthesis
 - synthesis
- pH 7
- c
- One of: enzymes, cell structure, membrane transport, muscle
- A molecule is a substance composed of two or more atoms that are bonded together covalently. A macromolecule is a large, complex molecule. Usually, the subunits of a macromolecule are small molecules bonded together.
- An enzyme is a macromolecule that speeds up chemical reactions.
- c
- c
- b

Review Unit 1 Biochemistry Answers

M Mosston



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Stephenson, 1978. A comprehensive up to date treatment of the biochemistry essential for an understanding of molecular and cellular biological processes. This third edition offers new units covering the chemistry of life, bioenergetics, energy transfer.

molecules regulation of enzymes and reaction sequences lab techniques for purification of proteins and nucleic acids and lab techniques of molecular genetics Also each unit contains more applications to biological systems The text provides a well organized and rigorous approach suitable for classroom use or self instruction Each unit begins with a 1 to 2 page presentation of basic concepts followed by about 20 questions and problems with sample responses Self tests appear after every 2 to 3 units and there is a cumulative self test at the end of the book **Nuclear Science Abstracts** ,1973-07

Research Anthology on Developing Critical Thinking Skills in Students Management Association, Information Resources,2020-10-02 Learning strategies for critical thinking are a vital part of today s curriculum as students have few additional opportunities to learn these skills outside of school environments Therefore it is essential that educators be given practical strategies for improving their critical thinking skills as well as methods to effectively provide critical thinking skills to their students The Research Anthology on Developing Critical Thinking Skills in Students is a vital reference source that helps to shift and advance the debate on how critical thinking should be taught and offers insights into the significance of critical thinking and its effective integration as a cornerstone of the educational system Highlighting a range of topics such as discourse analysis skill assessment and measurement and critical analysis techniques this multi volume book is ideally designed for teachers instructors instructional designers curriculum developers education professionals administrators policymakers researchers and academicians **Digital Technologies and Instructional Design for Personalized Learning** Zheng, Robert,2018-03-02 When facilitating high quality education using digital technology to personalize students learning is a focus in the development of instruction There is a need to unify the multifaceted directions in personalized learning by presenting a coherent and organized vision in the design of personalized learning using digital technology Digital Technologies and Instructional Design for Personalized Learning is a critical scholarly resource that highlights the theories principles and learning strategies in personalized learning with digital technology Featuring coverage on a broad range of topics such as collaborative learning instructional design and computer supported collaborative learning this book is geared towards educators professionals school administrators academicians researchers and students seeking current research on the area of personalized learning with digital technology DNA Crime Labs United States. Congress. Senate. Committee on the Judiciary,2002 **Self-assessment of Current Knowledge in Urology** Stephen M. Lazarus,1972 National Library of Medicine AVLINE Catalog National Library of Medicine (U.S.),1975 Listing of audiovisual materials catalogued by NLM Items listed were reviewed under the auspices of the American Association of Dental Schools and the Association of American Medical Colleges and are considered suitable for instruction Entries arranged under MeSH subject headings Entry gives full descriptive information and source Also includes Procurement source section that gives addresses and telephone numbers of all sources **Next Generation Level 2 Student's Book** Eoin Higgins,2012-06-30 Student s Book 2 has six units In each unit a Get started section helps students with strategies needed for effective language learning Grammar

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