

Fill in the blank with the correct word or phrase.

1. The theory that explains the behavior of gases at the molecular level is called the _____, which is based on assumptions about the _____ of the gas molecules.
2. Use the kinetic theory to explain why a helium-filled balloon expands when it is brought inside a warm room from the cold outdoors.

3. Define "RMS speed".

4. Gases deviate more from ideal behavior under _____ conditions.

5. Pressure is defined as _____ per unit _____. The earth's atmosphere has weight, which creates _____.
6. Standard temperature has been established as _____ °C or _____ K. Standard pressure has been established as _____ atm, _____ mm Hg, or _____ Pa. The difference between standard temperature and pressure is _____.
7. At _____, the total pressure atmospheric pressure, which is _____, is equal to molecular gas pressure.
8. When atmospheric pressure increases, how does the height of the column of mercury change?

9. If pressure is constant, the volume of a sample of gas _____ if temperature increases.
10. At constant pressure, the volume of a sample of gas is _____ if temperature is increased or _____ if temperature decreases.

Review Sheet Unit 9 Chemistry Answers

IM Harris



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