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Chemistry - Unit 7 Review

Chemical Reaction Model

Describe key characteristics of all chemical reactions, including the rule of energy.
 Explain how a balanced equation represents these features unclude an enough).

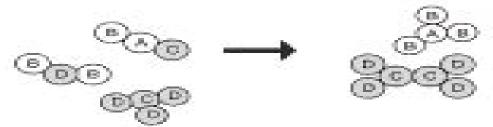
In chemical exactions, atoms of the reactants recombine to from new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating 600 between the system and sucroundings. The bolimaced chemical equation represents this process symbolically, showing that the atoms in the products are the same as these in the reactions.

2 H_s + O_{v→} 2 H_sO + energy

Note: there are 4 atoms of H and 2 atoms of O an each side of the equation.

Applying the Model

Balance the reaction below by drawing any additional reaction and/or product molecules that would be needed.



Write the balanced equation for this reaction.

$$DB_2 + CD_3 + BAC \rightarrow C_2D_4 + AB_6$$

Which of these nulecules would be called reactants?

Review Sheet Unit 7 Answers

B Lingard

Review Sheet Unit 7 Answers:

Platform One Pat Black,2000 This assessment and review pack is designed to assess children's learning at the end of each half term. The activities provided are linked directly to the Key Learning Objectives outlined in the Framework for Teaching Mathematics and the new NNS guidelines for assessment and review lessons. Alfred's Piano 101: Teacher's Handbook for Books 1 & 2 E. L. Lancaster, Kenon D. Renfrow, The Teacher's Handbook serves as an aid in curriculum development and daily lesson planning Suggested daily lesson plans suggested assignments following each lesson plan teaching tips for each unit suggested examinations for the semester and answer keys for the written exercises and unit review worksheets. It also suggests ways to successfully integrate keyboard and computer technology into the curriculum

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