



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

DATE: / /
TIME: hours



The question paper consists of 10 pages, including 4 extra sheets.
Use blue or black ink for writing. Do not use correction fluid.

QUESTION 3

PHYSICAL SCIENCES PAPER 2

INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

Examiner's use: _____ Physical Sciences P2

QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the ΔH_{vap} and ΔH_{fusion} for 1-hexanol in kJ mol⁻¹ and 1 mol of 1-hexanol.

Compound	Molecular mass (M _r)	Boiling point (°C)
A	102	16,2
B	102	21

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol only composed of A and B? at 0 °C and 101 kPa of external pressure? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of both compounds A and B and its relative value of these molecules. (2)
- 3.1.4 What (100%) of the compound A and B is 1-hexanol? Explain how you arrived at your answer. (2)
- 3.1.5 Which compound (A or B) will have the higher vapour pressure? (2)
- 3.1.6 Give a reason for your answer. (2)
- 3.1.7 Which compound (A or B) will have the higher surface tension? (2)
- 3.1.8 Give a reason for the answer to 3.1.7 referring to the space for 3.1.6. (2)

Compound C (M_r 102) will also be in the table. Use a suitable intermolecular force that both compounds A and B do not have to identify liquid boiling point of 16,2 °C.

- 3.1.8 Explain why the boiling point of compound C is 100 °C (100 °C) based on the intermolecular force that it has and its relative value of these molecules. (2)

3.2 Add table 3.2 to table 3.1 and use the values of ΔH_{vap} and ΔH_{fusion} to identify the liquid in the diagram below.



- 3.2.1 Is there liquid (H₂O) or H₂O, at the water level? (2)
- 3.2.2 Explain the answer to QUESTION 3.2.1 about the relative value of the intermolecular forces involved. (2)

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