

Note Taking Guide: Episode 301 Name_____

Model: A familiar idea used to explain unfamiliar facts observed in nature.

Theory: An explanation of observable facts and phenomena. To

remain valid, models and theories must:

- explain all known facts
- enable scientists to make correct predictions.

Democritus:

- proposed the existence of an atom
- word comes from the greek word atomis which means not to cut or "indivisible"

Aristotle

- rejected the idea of the atom
- said matter could be cut continually

Dalton's theory proposed that atoms:

- are the building blocks of matter
- are indivisible
- of the same elements are identical
- of different elements are different
- unite in small, whole number ratios to form compounds.

J.J. Thomson

- credited with discovery of electron: a blow to Dalton's indivisible atom.
- proposed the plum pudding model of the atom:
Negatively charged electrons embedded in a ball of positive charge

Note Taking Guide Episode 301 Chemistry

WJ Hussar



Note Taking Guide Episode 301 Chemistry:

Unveiling the Magic of Words: A Review of "**Note Taking Guide Episode 301 Chemistry**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Note Taking Guide Episode 301 Chemistry**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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