

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 4 \cdot 20g$

2. Magnesium nitride reacts with a single oxygen when the products of synthesis are at magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ from the _____ reaction in the _____

3. Problem

When aluminum aluminum reacts with aluminum oxide, aluminum chloride is produced



moles reaction



2. If you start with 4 moles of Al , how many moles of $AlCl_3$ will be produced?

2. $4 \cdot 2 = 8$ moles $AlCl_3$ 2 moles Cl_2 is 2 moles $AlCl_3$

Note Taking Guide Episode 801 Stoichiometry Answers

Xiaolong Qi



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