

### Atomic Number

- Increase across a period
- increase down a family

### Atomic Mass

- increase across a period
- increase down a family

### Atomic Radius

half the distance between nuclei of two like atoms.

- decreases across a period (due to increased attraction of more positive nuclei to electron clouds)
- increase down a family (due to addition of an energy level)

### Electron Affinity

amount of energy when an atom gains an electron

- increases across a period
- decreases down a family

### Ionization Energy

amount of energy required to remove an electron from a neutral atom

- increase across a period
- decrease down a family

### Electron Distribution - Forming Ions

- Atoms become ions by gaining or losing electrons to become an ion.
- An atom is stable when it has a full outer energy level (8 valence electrons).
- Atoms that lose (give away) electrons become positive charged ions. Atoms that gain (steal) electrons become negative charged ions.

# Note Taking Guide Episode 403

**Jiyuan Zhang**



## **Note Taking Guide Episode 403:**

## **Note Taking Guide Episode 403** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Note Taking Guide Episode 403**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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