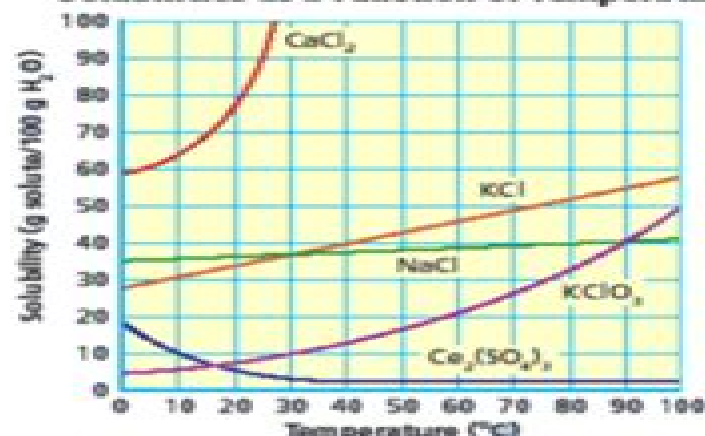


Name: _____

Date: _____ Period: _____

Solubility Curve Worksheet 2Directions: Use the graph below to answer the following questions.**Solubilities as a Function of Temperature****Types of Solutions**

- Points that collectively make up the solubility curve (points **ON** the curve) represent _____ solutions.
- Points that are **BELOW** the curve represent _____ solutions.
- Points **ABOVE** the solubility curve represent _____ solutions, and the **difference** between the point above the curve and on the curve represents the amount of solute which will precipitate out.
- Say if the resulting solutions would be saturated, supersaturated or unsaturated.
 - 60 g of KCl at 70°C _____
 - 10 g of KClO₃ at 60°C _____
 - 70 g of CaCl₂ at 20°C _____

1. a. What is the solubility of calcium chloride (CaCl₂) at 5°C? _____
 b. What is the solubility of calcium chloride at 25°C? _____
 c. What is happening to the solubility and temperature for CaCl₂? Solubility is _____ as temperature is _____.
2. a. What is the solubility of cerium sulfate - Ce₂(SO₄)₃ at 10°C? _____
 b. What is the solubility of cerium sulfate at 50°C? _____
 c. What is happening to the solubility and temperature for Ce₂(SO₄)₃? Solubility is _____ as temperature is _____.
3. Which substance on the graph is **least** soluble at 10°C? _____
4. Which substance on the graph shows the **least** change in solubility from 0°C to 100°C? _____
5. Which solution is more concentrated:
 - a) At 10°C, a saturated solution of NaCl or a saturated solution of CaCl₂ (circle one)
 - b) At 55°C, a saturated solution of KClO₃ or a saturated solution of Ce₂(SO₄)₃ (circle one)
6. What is the mass of cerium sulfate that will dissolve in 50 g of water at 10 degrees Celsius?
7. What is the mass of potassium chloride that will dissolve in 250 g of water at 70 degrees Celsius?
8. At 90 degrees Celsius, 10 g of potassium chlorate is dissolved in 100 g of water. Is this solution saturated, unsaturated, or supersaturated? How do you know?
9. A saturated solution of potassium chlorate is dissolved in 100g of water. If the saturated solution is cooled from 90 degrees Celsius to 60 degrees Celsius, how many grams would crystallize out?

Quiz Solubility Graphs Answer Key

Mike Jess

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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Table of Contents Quiz Solubility Graphs Answer Key

1. Understanding the eBook Quiz Solubility Graphs Answer Key
 - The Rise of Digital Reading Quiz Solubility Graphs Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Quiz Solubility Graphs Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Quiz Solubility Graphs Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Quiz Solubility Graphs Answer Key
 - Personalized Recommendations
 - Quiz Solubility Graphs Answer Key User Reviews and Ratings
 - Quiz Solubility Graphs Answer Key and Bestseller Lists
5. Accessing Quiz Solubility Graphs Answer Key Free and Paid eBooks
 - Quiz Solubility Graphs Answer Key Public Domain eBooks
 - Quiz Solubility Graphs Answer Key eBook Subscription Services
 - Quiz Solubility Graphs Answer Key Budget-Friendly Options

6. Navigating Quiz Solubility Graphs Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Quiz Solubility Graphs Answer Key Compatibility with Devices
 - Quiz Solubility Graphs Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Quiz Solubility Graphs Answer Key
 - Highlighting and Note-Taking Quiz Solubility Graphs Answer Key
 - Interactive Elements Quiz Solubility Graphs Answer Key
8. Staying Engaged with Quiz Solubility Graphs Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Quiz Solubility Graphs Answer Key
9. Balancing eBooks and Physical Books Quiz Solubility Graphs Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Quiz Solubility Graphs Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Quiz Solubility Graphs Answer Key
 - Setting Reading Goals Quiz Solubility Graphs Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Quiz Solubility Graphs Answer Key
 - Fact-Checking eBook Content of Quiz Solubility Graphs Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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