

Unit 1: Matter and Energy

Unit Review

Name: **KEY**

Date: _____ Class: _____

Directions: Complete the following review questions. Review your Cornell notes, learning targets, worksheets, labs, and other assignments to adequately prepare for the unit test.

Learning Target(s): 1, 2, 3, 4, 10

1. In the Melting of a Solid lab, you heated (or cooled) an unknown solid and recorded its temperature at 30-second intervals. From this data you created a heating (or cooling) curve to identify the substance.
 - a. Identify the independent variable of this experiment: **the time you heat the solid for**
 - b. Identify the dependent variable of this experiment: **the temperature of the solid**
 - c. Identify the control for this experiment: **substance used; method of heating**
2. Decide whether the following observations are qualitative or quantitative:
 - a. The solution began to get cloudy once the second reagent was added. **Qualitative**
 - b. The piece of magnesium was 1.25 centimeters in length. **Quantitative**
 - c. The mass of the block decreased by fifteen grams. **Quantitative**
 - d. The test tube appeared to have a dark orange solid suspended in the solution. **Qualitative**
 - e. The residue left in test tube 5 was slightly larger than in test tube 4. **Qualitative**

Learning Target(s): 1-15

3. Suggest a process to use to separate each of the following mixtures:
 - a. A mix of salt and water: _____
 - b. A smoothie drink with veggies and fruits of different densities: _____
4. Which physical property does filtration use to separate a mixture? _____

Learning Target(s): 5, 6

5. Describe at least two safety precautions to use when using a Bunsen Burner in lab:
 - a. **Tie back hair and loose clothing**
 - b. **Light match first then turn on gas valve**
6. Which type of lab equipment would be most appropriate to measure 4 mL of water in? Choose one and explain why.

a. A 250 mL beaker b. A 10 mL graduated cylinder c. An electronic balance	d. A 50 mL graduated cylinder e. A large test tube f. A thermometer
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You would get the most accurate measurement with this size. This piece of equipment has graduations to measure volume.

Physical Science Chemistry Unit Review Guide Answers

Brendan G. Carr



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Table of Contents Physical Science Chemistry Unit Review Guide Answers

1. Understanding the eBook Physical Science Chemistry Unit Review Guide Answers
 - The Rise of Digital Reading Physical Science Chemistry Unit Review Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chemistry Unit Review Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chemistry Unit Review Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chemistry Unit Review Guide Answers

- Personalized Recommendations
 - Physical Science Chemistry Unit Review Guide Answers User Reviews and Ratings
 - Physical Science Chemistry Unit Review Guide Answers and Bestseller Lists
5. Accessing Physical Science Chemistry Unit Review Guide Answers Free and Paid eBooks
 - Physical Science Chemistry Unit Review Guide Answers Public Domain eBooks
 - Physical Science Chemistry Unit Review Guide Answers eBook Subscription Services
 - Physical Science Chemistry Unit Review Guide Answers Budget-Friendly Options
 6. Navigating Physical Science Chemistry Unit Review Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
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 - Physical Science Chemistry Unit Review Guide Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chemistry Unit Review Guide Answers
 - Highlighting and Note-Taking Physical Science Chemistry Unit Review Guide Answers
 - Interactive Elements Physical Science Chemistry Unit Review Guide Answers
 8. Staying Engaged with Physical Science Chemistry Unit Review Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chemistry Unit Review Guide Answers
 9. Balancing eBooks and Physical Books Physical Science Chemistry Unit Review Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chemistry Unit Review Guide Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science Chemistry Unit Review Guide Answers
 - Setting Reading Goals Physical Science Chemistry Unit Review Guide Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science Chemistry Unit Review Guide Answers

- Fact-Checking eBook Content of Physical Science Chemistry Unit Review Guide Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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