#### Chapter 5 Study Guide- Physical Science Test Monday, April 28<sup>th</sup>, 2014

- What is a chemical bond? force of attraction that holds elements together in order to form a unit
- Define chemical structure, the arrangement of atoms in a substance.
- What is an ionic bond? a chemical bond in which one atom loses an electron to form a positive ion and the other atom gains an electron to form a negative ion
- 4. An ionic bond is between a METAL and a NONMETAL .
- What is a covalent bond? a chemical bond that involves sharing a pair of electrons between atoms in a molecule
- 6. A covalent bond is between a nonmetal and a nonmetal .
- What is a metallic bond? a bond formed by the attraction between positively charged metal ions and the electrons around them.
- 8. What are the polyatomic ions? an ion made of two or more atoms.
- What is an ION? a charged atom
- 10. What is a CATION? A positively charged atom
- 11. What is an ANION? A negatively charged atom
- 12. Which part of an ionic compound should be written first? Metal

#### Name the following ionic compounds:

- 13. Fel2- Iron Iodide
- 14. MnF3- Manganese Fluoride
- 15. CrCl2- Chromium Chloride
- 16. CuS- Copper Sulfide

#### Name the following covalent compounds:

- 17. As2Os- diarsenic pentaoxide
- 18. Sils- silicon tetrafodide
- 19. P4S3- tertraphosphorus trisulfide
- 20. PaOin- tetraphosphorus decoxide

#### Write the equation for the following ionic compounds:

- 21. Lithium Oxide- LizO
- 22. Magnesium Nitride- Mg3N2
- Sodium Phosphide- NasP
- 24. Calcium Bromide CaBrz

#### Define/Explain the following

- 25. What is a subscript? number written below, tells you how many of each element is present in compound/molecule
- 26. What is a superscript? number written above the element symbol, this is how the oxidation number (charge) is displayed.

# **Physical Science Chemical Bond Study Guide**

SJ Ball

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Standards A Framework for Research in Mathematics Science and Technology Education National Research Council 2002
That report provided organizing principles for the design conduct and interpretation of research regarding the influence of national standards The Framework developed in Phase I was used to structure the current review of research that is reported here Phase II began in mid 2001 involved a thorough search and review of the research literature on the influence of the NSES and concludes with this publication which summarizes the proceedings of a workshop conducted on May 10 2002 in Washington DC Phase III will provide input collected in 2002 from science educators administrators at all levels and other practitioners and policy makers regarding their views of the NSES the ways and extent to which the NSES are influencing their work and the systems that support science education and what next steps are needed

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## **Table of Contents Physical Science Chemical Bond Study Guide**

- 1. Understanding the eBook Physical Science Chemical Bond Study Guide
  - The Rise of Digital Reading Physical Science Chemical Bond Study Guide
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Chemical Bond Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Chemical Bond Study Guide
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Chemical Bond Study Guide
  - Personalized Recommendations
  - Physical Science Chemical Bond Study Guide User Reviews and Ratings
  - Physical Science Chemical Bond Study Guide and Bestseller Lists
- 5. Accessing Physical Science Chemical Bond Study Guide Free and Paid eBooks
  - Physical Science Chemical Bond Study Guide Public Domain eBooks
  - Physical Science Chemical Bond Study Guide eBook Subscription Services
  - Physical Science Chemical Bond Study Guide Budget-Friendly Options
- 6. Navigating Physical Science Chemical Bond Study Guide eBook Formats

- o ePub, PDF, MOBI, and More
- Physical Science Chemical Bond Study Guide Compatibility with Devices
- Physical Science Chemical Bond Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Physical Science Chemical Bond Study Guide
  - Highlighting and Note-Taking Physical Science Chemical Bond Study Guide
  - o Interactive Elements Physical Science Chemical Bond Study Guide
- 8. Staying Engaged with Physical Science Chemical Bond Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Chemical Bond Study Guide
- 9. Balancing eBooks and Physical Books Physical Science Chemical Bond Study Guide
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Physical Science Chemical Bond Study Guide
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Chemical Bond Study Guide
  - Setting Reading Goals Physical Science Chemical Bond Study Guide
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Chemical Bond Study Guide
  - Fact-Checking eBook Content of Physical Science Chemical Bond Study Guide
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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